

Oxidative Aromatization, Cytotoxic Activity Evaluation and Conformational Study of Novel 7-aryl-10, 11-dihydro-7H-chromeno [4, 3-b]quinoline-6, 8(9H, 12H)-dione Derivatives

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Abstract

In the present work, novel 7-aryl-10, 11-dihydro-7H-chromeno [4, 3-b]quinoline-6, 8(9H, 12H)-dione derivatives were synthesized by oxidation of 7-aryl-8, 9, 10, 12-tetrahydro-7H-chromeno[4, 3-b]quinoline-6, 8-diones in the presence of silica sulfuric acid/NaNO₂ with yields of 64-74%. Cytotoxic activity of synthesized compounds was assessed on three different human cancer cell lines (K562, LS180, and MCF-7). Synthesized compounds showed moderate cytotoxic activities. The most active one appeared to be 2e, containing a methoxy group on the meta position of phenyl ring (IC₅₀ range in different cell lines: 11.1–55.7 μM). Furthermore; comparison of the cytotoxic activity of these novel oxidized derivatives with non-oxidized counterparts revealed that oxidation of dihydropyridine ring to pyridine, improves the activity especially in LS180 cell line. Conformational analysis revealed that some conformational aspects of oxidized derivatives such as orientation of C₇-aryl substitute were clearly different from non-oxidized ones.

Keywords: Oxidative aromatization; Chromeno [4, 3-b]quinoline; Cytotoxicity; Conformational analysis.

Introduction

1, 4-Dihydropyridine (DHP) compounds are well known as calcium channel modulators and have emerged as one of the most important classes of drugs for the treatment of cardiovascular diseases (1). Although the DHP nucleus has been particularly well explored as L-type calcium channel modulator, DHP is a privileged structure or scaffold that can, when appropriately decorated, interact with diverse

receptors and possess a variety of biological activities (2).

Previous studies have demonstrated the cytotoxic and anticancer activity of some 1, 4-DHPs derivatives (3-5). Moreover, the results of different studies indicate that 1, 4-DHP derivatives have significant inhibitory effects on MDR in cancer cell lines (6-8).

On the other hand, synthetic and naturally occurring coumarine derivatives are one of the most promising scaffolds in medicinal chemistry (9-11). In addition, different coumarine derivatives such as 4-hydroxycoumarine and 7-hydroxycoumarine derivatives demonstrated

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cytotoxic and antitumoral properties (12-13).

We have previously synthesized and evaluated the cytotoxic activity of some novel heteroanalogues of fused DHPs with the features of 1,4-DHPs and 4-hydroxycoumarins named as “chromeno[4, 3-*b*]quinoline or 7-aryl-8,9,10,12-tetrahydro-7*H*-chromeno[4, 3-*b*]quinoline-6,11-dione derivatives” 1 (14). Some of these derivatives showed moderate cytotoxic capacity and at the same time very low calcium channel antagonist activity, an undesirable effect when these compounds are used as antitumoral agents. These findings prompted us to further optimize this structure for design of more potent and specific cytotoxic agents. Armed with our experience and our interest in pharmacological properties and especially cytotoxic and DNA-intercalating activity of polycondensed heterocyclic compounds such as 1,8-acridinone and benzopyrano[3, 2-*c*]chromene-6, 8-dione derivatives (15-18), we were persuaded to aromatize these newly synthesized 1,4-DHPs (1) in order to obtain new pyridine derivatives with higher cytotoxic effect. We also examined some structural properties by means of computational conformational analysis.

Experimental

Chemistry

All chemicals and solvents used in this study were purchased from Merck (Darmstadt, Germany) and Sigma-Aldrich (San Louis, MO, USA). Melting points were determined on a Kofler hot stage apparatus and are uncorrected. The IR spectra were obtained on a Shimadzu 470 spectrophotometer (potassium bromide disks).

¹H-NMR spectra were measured using a Bruker FT-500 spectrometer, and chemical shifts are expressed as δ (ppm) with tetramethylsilane as internal standard. The mass spectra were run on a Finnigan TSQ-70 spectrometer at 70 eV. Merck silica gel 60 F254 plates were used for analytical TLC; column chromatography was performed on Merck silica gel (70–230 mesh). Yields were calculated for purified products and were not optimized.

*Typical Procedure for synthesis of 7-Aryl-10, 11-dihydro-9*H*-chromeno[4,3-*b*]quinoline-6,8-dione (2)*

Compounds 1 (5.4 mmol), sodium nitrite (16 mmol, 1.1 g) and silica sulfuric acid (1.6 g) and silica (1.07 g) were refluxed in chloroform (100 mL) for 8-12 h. The mixture was cooled to room temperature and was filtered. The filtrate was evaporated to dryness under reduced pressure, and the crude product was purified by short column chromatography to give 2 (Table 1).

*7-(2-Methylphenyl)-10, 11-dihydro-6*H*-chromeno[4, 3-*b*]quinoline-6, 8-dione (2a).*

Yellowish solid 1.36 g (yield 71%), mp = 210-212 °C, $R_f = 0.70$ (2:8 EtOAc/Petroleum Ether).

IR ν_{\max} cm^{-1} (KBr): 2922, 2848 (C-H aliphatic), 1754 (C = O ester), 1691 (C = O ketone), 757 (C-H bending).

¹HNMR (CDCl₃), δ (ppm): 2.02 (s, 3H, CH₃), 2.24 (qn, 2H, $J = 6.6$ Hz, H₁₀), 2.66 (t, 2H, $J = 6.6$ Hz, H₉), 3.38 (t, 2H, $J = 6.6$ Hz, H₁₁), 6.81 (d, ¹H, $J = 7.0$ Hz, H₁₅), 7.23 (t, ¹H, $J = 7.0$ Hz, H₁₇), 7.28-7.35 (m, 3H, H₁₆, H₁₈, H₄), 7.40 (t, ¹H, $J = 8.0$ Hz, H₂), 7.61 (t, ¹H, $J = 8.0$ Hz, H₃), 8.68 (d, ¹H, $J = 8.0$ Hz, H₁).

¹³CNMR (CDCl₃), δ: 20.01, 20.98, 34.61, 40.13, 115.13, 116.90, 118.83, 124.61, 124.89, 125.58, 126.09, 127.44, 127.69, 129.26, 133.33, 134.24, 137.84, 153.43, 154.13, 156.39, 157.92, 169.58, 196.14.

MS : m/z (%), 355 (M⁺, 44), 340 (44), 327 (37), 299(100), 271(37), 151(50), 126 (50), 114 (63), 100 (44), 87 (25).

Anal. Calcd for C₂₃H₁₇NO₃: C, 77.73; H, 4.82; N, 3.94. Found: C, 77.40 ; H, 4.56; N, 3.95.

*7-(3-Methylphenyl)-10, 11-dihydro-6*H*-chromeno[4, 3-*b*]quinoline-6, 8-dione (2b).*

White solid, 1.22 g (yield 64%), mp = 172-174 °C, $R_f = 0.73$ (2:8 EtOAc/Petroleum Ether).

IR ν_{\max} cm^{-1} (KBr): 2848 (C-H aliphatic), 1754 (C=O ester), 1691(C=O ketone), 757 (C-H bending).

¹HNMR (CDCl₃), δ (ppm): 2.24 (qn, 2H, $J = 6.6$ Hz, H₁₀), 2.39 (s, 3H, CH₃), 2.67 (t, 2H, $J = 6.5$ Hz, H₉), 3.36 (t, 2H, $J = 6.5$ Hz, H₁₁), 6.91 (m, 2H, H₁₄, H₁₆), 7.24 (d, ¹H, $J = 7.5$ Hz, H₁), 7.30 (d, ¹H, $J = 8.0$ Hz, H₄), 7.33 (t, ¹H, $J = 7.5$ Hz, H₁₇), 7.39 (t, ¹H, $J = 8.0$ Hz, H₂), 7.60 (t, ¹H, $J =$

8.0 Hz, H₃), 8.67 (d, ¹H, *J* = 8.0 Hz, H₁).

¹³CNMR (CDCl₃), δ: 20.93, 21.64, 34.56, 40.33, 115.08, 116.83, 118.80, 123.34, 124.56, 125.67, 126.11, 126.78, 127.74, 128.41, 133.25, 133.40, 137.35, 153.41, 153.86, 156.60, 158.13, 169.29, 196.39.

MS : *m/z* (%), 355 (M⁺, 48), 340 (38), 265 (38), 149 (86), 121 (100), 105 (23), 92 (35), 71 (35), 57 (42), 43 (50).

Anal. Calcd for C₂₃H₁₇NO₃: C, 77.73; H, 4.82; N, 3.94. Found: C, 77.80; H, 4.85; N, 3.97.

7-(4-Methylphenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2c).

White solid, 1.30 g (yield 68%) mp = 241-243 °C, R_f = 0.74 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm⁻¹(KBr): 3033 (C-H aromatic), 2853 (C-H aliphatic), 1750 (C=O ester), 1698 (C=O ketone), 759 (C-H bending).

¹HNMR (CDCl₃), δ (ppm): 2.24 (qn, 2H, *J* = 6.6 Hz, H₁₀), 2.46 (s, 3H, CH₃), 2.68 (t, 2H, *J* = 6.6 Hz, H₉), 3.37 (t, 2H, *J* = 6.3 Hz, H₁₁), 7.02 (d, 2H, *J* = 8.0 Hz, H₁₅, H₁₇), 7.27 (d, 2H, *J* = 8.0 Hz, H₁₄, H₁₈), 7.32 (d, ¹H, *J* = 8.2 Hz, H₄), 7.41 (t, ¹H, *J* = 8.2 Hz, H₂), 7.61 (t, ¹H, *J* = 8.2 Hz, H₃), 8.68 (d, ¹H, *J* = 8.2 Hz, H₁).

¹³CNMR (CDCl₃), δ: 20.94, 21.57, 34.54, 40.33, 115.19, 116.86, 118.84, 124.56, 126, 11, 126.15, 127.94, 128.75, 133.24, 134.93, 137.19, 153.42, 153.89, 156.67, 158.25, 169.27, 196.58.

MS : *m/z* (%), 355 (M⁺, 100), 340 (31), 327 (25), 299 (19), 127 (13).

Anal. Calcd for C₂₃H₁₇NO₃: C, 77.73; H, 4.82; N, 3.94. Found: C, 77.68; H, 4.78; N, 3.90.

7-(2-Methoxyphenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione(2d).

White solid, 1.42 g (yield 71%), mp = 180-182 °C, R_f = 0.69 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm⁻¹(KBr): 3056 (C-H aromatic), 2852 (C-H aliphatic), 1749 (C=O ester), 1693 (C=O ketone), 753 (C-H bending).

¹HNMR (CDCl₃), δ (ppm): 2.25 (qn, 2H, *J* = 6.7 Hz, H₁₀), 2.69 (t, 2H, *J* = 6.7 Hz, H₉), 3.38 (t, 2H, *J* = 6.7 Hz, H₁₁), 3.74 (s, 3H, OCH₃), 6.90 (d, ¹H, *J* = 6.0 Hz, H₁₅), 6.89 (d, ¹H, *J* = 6.0 Hz, H₁₈), 7.00 (m, ¹H, H₁₇), 7.31 (d, ¹H, *J* = 8.2 Hz, H₄), 7.41 (t, ¹H, *J* = 6.0 Hz, H₁₆), 7.43 (t, ¹H, *J* = 8.2 Hz, H₂), 7.61 (t, ¹H, *J* = 8.2 Hz, H₃), 8.68 (d, ¹H, *J* = 8.2 Hz, H₁).

¹³CNMR (CDCl₃), δ: 20.97, 34.48, 40.06, 55.67, 110.39, 115.60, 116.82, 118.97, 120.73, 124.49, 126.04, 126.76, 127.40, 128.07, 129.21, 133.08, 153.32, 153.51, 153.87, 155.85, 158.13, 169.18, 196.28.

MS : *m/z* (%), 371 (M⁺, 38), 340 (100), 120 (31), 91 (31), 75 (25).

Anal. Calcd for C₂₃H₁₇NO₄: C, 74.38; H, 4.61; N, 3.77. Found: C, 74.32; H, 4.58; N, 3.70.

7-(3-Methoxyphenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2e).

White solid, 1.32 g (yield 66%), mp = 155-157 °C, R_f = 0.67 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm⁻¹(KBr): 3060 (C-H aromatic), 2835 (C-H aliphatic), 1755 (C=O ester), 1692 (C=O ketone), 766 (C-H bending).

¹HNMR (CDCl₃), δ (ppm): 2.25 (qn, 2H, *J* = 6.5 Hz, H₁₀), 2.68 (t, 2H, *J* = 6.5 Hz, H₉), 3.37 (t, 2H, *J* = 6.5 Hz, H₁₁), 3.82 (s, 3H, OCH₃), 6.67 (s, ¹H, H₁₄), 6.72 (d, ¹H, *J* = 7.5 Hz, H₁₈), 6.98 (d, ¹H, *J* = 7.5 Hz, H₁₆), 7.31 (d, ¹H, *J* = 8.0 Hz, H₄), 7.37-7.42 (m, 2H, H₁₇, H₂), 7.61 (t, ¹H, *J* = 8.0 Hz, H₃), 8.68 (d, ¹H, *J* = 8.0 Hz, H₁).

¹³CNMR (CDCl₃), δ: 20.91, 34.56, 40.29, 55.13, 112.49, 112.56, 115.02, 116.86, 118.76, 124.57, 126.11, 127.62, 128.97, 133.29, 133.89, 139.26, 153.41, 153.91, 155.96, 157.97, 159.33, 169.35, 196.17.

MS : *m/z* (%), 371 (M⁺, 44), 340 (37), 341 (50), 196 (63), 120 (38), 91 (100), 43 (50).

Anal. Calcd for C₂₃H₁₇NO₄: C, 74.38; H, 4.61; N, 3.77. Found: C, 74.33; H, 4.56; N, 3.68.

7-(4-Methoxyphenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione(2f).

White solid, 1.50 g (yield 75%), mp = 185-187 °C, R_f = 0.71 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm⁻¹(KBr): 2854(C-H aliphatic), 1747(C=O ester), 1688(C=O ketone), 761(C-H bending).

¹HNMR (CDCl₃), δ (ppm): 2.24 (qn, 2H, *J* = 6.6 Hz H₁₀), 2.69 (t, 2H, *J* = 6.5 Hz, H₉), 3.37 (t, 2H, *J* = 6.4 Hz, H₁₁), 3.88 (s, 3H, OCH₃), 7.00 (d, 2H, *J* = 8.7 Hz, H₁₅, H₁₇), 7.05 (d, 2H, *J* = 8.7 Hz, H₁₄, H₁₈), 7.32 (d, ¹H, *J* = 8.2 Hz, H₄), 7.41 (t, ¹H, *J* = 8.2 Hz, H₂), 7.62 (t, ¹H, *J* = 8.2 Hz, H₃), 8.68 (d, ¹H, *J* = 8.2 Hz, H₁).

¹³CNMR (CDCl₃-d), δ: 20.94, 34.54, 40.38, 55.11, 113.51, 115.29, 116.84, 118.85, 124.56,

126.12, 127.73, 128.11, 129.89, 130.03, 133.23, 153.40, 153.89, 156.37, 158.33, 169.25, 196.70.

MS : m/z (%), 371 (M^+ , 11), 289 (18), 149 (29), 121 (32), 85 (63), 71 (88), 57 (100), 43 (95).

Anal. Calcd for $C_{23}H_{17}NO_4$: C, 74.38; H, 4.61; N, 3.77. Found: C, 74.43; H, 4.59; N, 3.69.

7-(2-Chlorophenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2g).

White solid, 1.33 g (yield 66%), mp = 253-255 °C, R_f = 0.65 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm^{-1} (KBr): 3068 (C-H aromatic), 2854 (C-H aliphatic), 1744 (C=O ester), 1690 (C=O ketone), 762 (C-H bending).

1H NMR ($CDCl_3$), δ (ppm): 2.27 (qn, 2H, J = 6.5 Hz, H_{10}), 2.69 (t, 2H, J = 6.5 Hz, H_9), 3.41 (t, 2H, J = 6.5 Hz, H_{11}), 7.03 (d, 1H , J = 7.8 Hz, H_{18}), 7.27-7.43 (m, 4H, H_2 , H_4 , H_{16} , H_{17}), 7.50 (d, 1H , J = 7.8 Hz, H_{15}), 7.62 (t, 1H , J = 7.8 Hz, H_3), 8.70 (d, 1H , J = 7.8 Hz, H_1).

^{13}C NMR ($CDCl_3$), δ : 20.88, 34.55, 39.89, 114.97, 116.94, 118.75, 124.68, 126.08, 126.68, 127.09, 127.18, 128.92, 130.77, 133.42, 133.90, 137.28, 152.92, 153.38, 154.26, 157.99, 169.69, 196.94.

MS : m/z (%), 377 (M^+ +2, 33), 375 (M^+ , 100), 340 (25), 187 (13), 127 (13), 113 (13), 100 (10), 87 (5).

Anal. Calcd for $C_{22}H_{14}ClNO_3$: C, 70.31; H, 3.75; N, 3.73. Found: C, 70.40; H, 3.69; N, 3.70.

7-(3-Chlorophenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2h).

White solid, 1.37 g (yield 68%), mp = 188-190 °C, R_f = 0.64 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm^{-1} (KBr): 3018 (C-H aromatic), 2853 (C-H aliphatic), 1744 (C=O ester), 1691 (C=O ketone), 757 (C-H bending).

1H NMR ($CDCl_3$), δ (ppm): 2.23-2.28 (qn, 2H, J = 6.5 Hz, H_{10}), 2.67-2.70 (t, 2H, J = 6.5 Hz, H_9), 3.39-3.40 (t, 2H, J = 6.5 Hz, H_{11}), 7.02 (d, 1H , J = 7.0 Hz, H_{16}), 7.10 (s, 1H , H_{14}), 7.30 (d, 1H , J = 8.0 Hz, H_4), 7.37-7.42 (m, 3H, H_2 , H_{17} , H_{18}), 7.62 (t, 1H , J = 8.0 Hz, H_3), 8.68 (d, 1H , J = 8.0 Hz, H_1).

^{13}C NMR ($CDCl_3$), δ : 20.85 (C_{10}), 34.57 (C_{11}), 40.24 (C_9), 114.86 (C_7), 116.88 (C_4), 118.64 (C_{1a}), 124.59 (C_{16}), 124.72 (C_2), 126.15 (C_1), 126.27 (C_{14}), 127.26 (C_{7a}), 127.62 (C_{17}), 129.16 (C_{18}), 133.49 (C_3), 133.90 (C_{13}), 139.79 (C_{6a}),

153.38 (C_{12a}), 154.04 (C_{4a}), 154.48 (C_{15}), 158.07 (C_6), 169.64 (C_{11a}), 196.08 (C_8).

MS : m/z (%), 377 (M^+ +2, 10), 375 (M^+ , 30), 340 (30), 319 (27), 289 (14), 227 (44), 127 (44), 120 (100), 100 (64), 87 (32), 74 (36).

Anal. Calcd for $C_{22}H_{14}ClNO_3$: C, 70.31; H, 3.75; N, 3.73. Found: C, 70.38; H, 3.56; N, 3.67.

7-(4-Chlorophenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2i).

White solid, 1.41 g (yield 70%), mp = 281-283 °C, R_f = 0.66 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm^{-1} (KBr): 2851 (C-H aliphatic), 1745 (C=O ester), 1691 (C=O ketone), 753 (C-H bending).

1H NMR ($CDCl_3$), δ (ppm): 2.25 (qn, 2H, J = 6.6 Hz H_{10}), 2.68 (t, 2H, J = 6.6 Hz, H_9), 3.38 (t, 2H, J = 6.6 Hz, H_{11}), 7.05 (d, 2H, J = 8.3 Hz, H_{14} , H_{18}), 7.33 (d, 1H , J = 8.2 Hz, H_4), 7.41 (d, 1H , J = 8.2 Hz, H_2), 7.43 (d, 2H, J = 8.3 Hz, H_{15} , H_{17}), 7.62 (t, 1H , J = 8.2 Hz, H_3), 8.68 (d, 1H , J = 8.2 Hz, H_1).

^{13}C NMR ($CDCl_3$), δ : 20.87, 34.56, 40.28, 114.85, 116.90, 118.62, 124.70, 126.16, 126.25, 127.64, 128.29, 133.47, 133.88, 136.46, 153.39, 154.07, 155.15, 158.08, 169.55, 196.34.

MS : m/z (%), 377 (M^+ +2, 10), 375 (M^+ , 29), 289 (100), 265 (58), 237 (64), 121 (100), 85 (44), 71 (59), 57 (80).

Anal. Calcd for $C_{22}H_{14}ClNO_3$: C, 70.31; H, 3.75; N, 3.73. Found: C, 70.28; H, 3.70; N, 3.70.

7-(3-Nitrophenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2j).

White solid, 1.45 g (yield 70%), mp = 161-163 °C, R_f = 0.64 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm^{-1} (KBr): 3086 (C-H aromatic), 2849 (C-H aliphatic), 1748 (C=O ester), 1693 (C=O ketone), 1543, 1343 (N=O nitro aryl), 765 (C-H bending).

1H NMR ($CDCl_3$), δ (ppm): 2.28 (qn, 2H, J = 6.6 Hz, H_{10}), 2.68 (t, 2H, J = 6.6 Hz, H_9), 3.39 (t, 2H, J = 6.6 Hz, H_{11}), 7.33 (d, 1H , J = 8.0 Hz, H_4), 7.42-7.48 (m, 2H, H_2 , H_{17}), 7.61-7.66 (m, 2H, H_3 , H_{18}), 7.98 (s, 1H , H_{14}), 8.31 (d, 1H , J = 7.75 Hz, H_{16}), 8.70 (d, 1H , J = 8.0 Hz, H_1).

^{13}C NMR ($CDCl_3$), δ : 20.78, 34.58, 40.16, 114.75, 116.94, 118.51, 123.21, 123.47, 124.94, 125.40, 126.25, 126.93, 128.82, 132.68, 133.77, 139.90, 148.09, 153.30, 154.34, 158.38, 170.06, 196.26.

MS : m/z (%), 386 (M^+ , 30), 289 (37), 167 (62.5), 149 (51), 121 (29), 77 (98), 43 (100).

Anal. Calcd for $C_{23}H_{17}NO_3$: C, 77.73; H, 4.82; N, 3.94. Found: C, 77.69; H, 4.80; N, 3.86.

7-(4-Nitrophenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2k).

White solid, 1.39 g (yield 67%), mp = 265-267 °C, R_f = 0.63 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm^{-1} (KBr): 3072 (C-H aromatic), 2951, 2849 (C-H aliphatic), 1741 (C=O ester), 1689 (C=O ketone), 1547, 1346 (N=O nitro aryl), 765 (C-H bending).

1H NMR ($CDCl_3$), δ (ppm): 2.25 (qn, 2H, J = 6.4 Hz, H_{10}), 2.68 (t, 2H, J = 6.4 Hz, H_9), 3.39 (t, 2H, J = 6.4 Hz, H_{11}), 7.30 (d, 2H, J = 8.6 Hz, H_{14} , H_{18}), 7.33 (d, 1H , J = 8.2 Hz, H_4), 7.43 (t, 1H , J = 8.0 Hz, H_2), 7.64 (t, 1H , J = 8.0 Hz, H_3), 8.31 (d, 2H, J = 8.6 Hz, H_{15} , H_{17}), 8.70 (d, 1H , J = 8.0 Hz, H_1).

^{13}C NMR($CDCl_3$ -d), δ : 20.79, 34.58, 40.12, 114.54, 116.98, 118.49, 123.44, 124.96, 126.26, 126.66, 127.15, 133.82, 145.80, 147.17, 153.35, 153.83, 154.35, 158.30, 170.04, 196.15.

MS : m/z (%), 386 (M^+ , 12.5), 289(34), 265(18), 121(80), 84(100), 57(75).

Anal. Calcd for $C_{23}H_{17}NO_3$: C, 77.73; H, 4.82; N, 3.94. Found: C, 77.80; H, 4.92; N, 3.89.

7-(3-Bromophenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2l).

White solid, 1.44 g (yield 64%), mp = 184-186 °C, R_f = 0.72 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm^{-1} (KBr): 3070 (C-H aromatic), 2954 (C-H aliphatic), 1748 (C=O ester), 1693 (C=O ketone), 768 (C-H bending).

1H NMR ($CDCl_3$), δ (ppm): 2.25 (qn, 2H, J = 6.5 Hz, H_{10}), 2.69 (t, 2H, J = 6.5 Hz, H_9), 3.39 (t, 2H, J = 6.5 Hz, H_{11}), 7.08 (d, 1H , J = 7.6 Hz, H_{16}), 7.24 (s, 1H , H_{14}), 7.32, 7.35 (m, 2H, H_4 , H_{17}), 7.41 (t, 1H , J = 8.0 Hz, H_2), 7.58 (d, 1H , J = 7.6 Hz, H_{18}), 7.64 (t, 1H , J = 8.0 Hz, H_3), 8.68 (d, 1H , J = 8.0 Hz, H_1).

^{13}C NMR ($CDCl_3$), δ : 20.85, 34.58, 40.24, 114.88, 116.91, 118.64, 122.04, 124.71, 125.03, 126.16, 127.22, 128.96, 129.38, 130.52, 133.50, 140.01, 153.40, 154.08, 154.40, 158.08, 169.61, 196.06.

MS : m/z (%), 421 (M^+ +2, 98), 419 (100), 340 (33), 312 (68), 169 (80), 155 (38), 127 (67),

113 (70), 100 (60), 87 (47).

Anal. Calcd for $C_{22}H_{14}BrNO_3$: C, 62.87; H, 3.36; N, 3.33. Found: C, 62.78; H, 3.29; N, 3.40.

7-(4-Bromophenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2m).

White solid, 1.47 g (yield 65%), mp = 279-281 °C R_f = 0.77 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm^{-1} (KBr): 3042 (C-H aromatic), 2922 (C-H aliphatic), 1745 (C=O ester), 1693 (C=O ketone), 771 (C-H bending).

1H NMR ($CDCl_3$), δ (ppm): 2.24 (qn, 2H, J = 6.5 Hz, H_{10}), 2.67 (t, 2H, J = 6.5 Hz, H_9), 3.38 (t, 2H, J = 6.4 Hz, H_{11}), 6.98 (d, 2H, J = 8.3 Hz, H_{14} , H_{18}), 7.32 (d, 1H , J = 7.0 Hz, H_4), 7.40 (t, 1H , J = 7.0 Hz, H_2), 7.58 (d, 2H, J = 8.3 Hz, H_{15} , H_{17}), 7.62 (t, 1H , J = 7.0 Hz, H_3), 8.67 (d, 1H , J = 7.0 Hz, H_1). ^{13}C NMR ($CDCl_3$), δ : 20.87, 34.56, 40.28, 114.92, 116.91, 118.68, 124.71, 126.16, 126.26, 127.40, 127.91, 131.19, 133.48, 136.99, 153.39, 154.08, 155.10, 158.23, 169.57, 196.34.

MS : m/z (%), 421 (M^+ +2, 11), 419 (M^+ , 11), 330 (13), 289 (100), 216 (22), 149 (20), 83 (31), 57 (76).

Anal. Calcd for $C_{22}H_{14}BrNO_3$: C, 62.87; H, 3.36; N, 3.33. Found: C, 62.79; H, 3.37; N, 3.40.

7-(Phenyl)-10, 11-dihydro-6H-chromeno[4, 3-b]quinoline-6, 8-dione (2n).

Yellowish solid, 1.36 g (yield 74%), mp = 121-123 °C, R_f = 0.79 (2:8 EtOAc/Petroleum Ether).

IR ν_{max} cm^{-1} (KBr): 3056 (C-H aromatic), 2949 (C-H aliphatic), 1744 (C=O ester), 1693 (C=O ketone), 776 (C-H bending).

1H NMR ($CDCl_3$), δ (ppm): 2.25 (qn, 2H, J = 6.6 Hz, H_{10}), 2.68 (t, 2H, J = 6.6 Hz, H_9), 3.38 (t, 2H, J = 6.6 Hz, H_{11}), 7.13 (m, 2H, H_{14} , H_{18}), 7.32 (d, 1H , J = 8.0 Hz, H_4), 7.42 (t, 1H , J = 8.0 Hz, H_2), 7.46 (m, 3H, H_{15} , H_{16} , H_{17}), 7.61 (t, 1H , J = 8.0 Hz, H_3), 8.68 (d, 1H , J = 8.0 Hz, H_1).

^{13}C NMR ($CDCl_3$), δ : 20.91, 34.57, 40.30, 115.06, 116.87, 118.80, 124.59, 124.75, 126.13, 126.16, 127.56, 127.92, 133.30, 138, 153.43, 153.95, 156.38, 158.19, 169.38, 196.36.

MS : m/z (%), 341 (M^+ , 81), 313 (69), 289 (38), 196 (44), 120 (75), 90 (100), 76 (44).

Anal. Calcd for $C_{22}H_{15}NO_3$: C, 77.41, H, 4.43; N, 4.10. Found: C, 77.39; H, 4.40; N, 4.09.

Pharmacology

RPMI 1640, fetal bovine serum (FBS), trypsin and phosphate buffered saline (PBS) were purchased from Biosera (Ringmer, UK). 3-(4, 5-Dimethylthiazol-2-yl)-2, 5-diphenyltetrazolium bromide (MTT) was obtained from Sigma-Aldrich (Saint Louis, MO, USA) and penicillin/streptomycin was purchased from Invitrogen (San Diego, CA, USA). Doxorubicin and dimethyl sulphoxide were obtained from EBEWE Pharma (Unterach, Austria) and Merck (Darmstadt, Germany), respectively.

Cell lines and maintenance of human cell lines

K562 (human chronic myelogenous leukemia), LS180 (human colon adenocarcinoma) and MCF-7 (human breast adenocarcinoma) cells were obtained from the National Cell Bank of Iran, Pasteur Institute, Tehran, Iran. All cell lines were maintained in RPMI 1640 supplemented with 10% FBS, and 100 units/mL penicillin-G and 100 µg/mL streptomycin. Cells were maintained at 37 °C in humidified air containing 5% CO₂ and were grown in monolayer cultures except for K562 cells, which were grown in suspension.

MTT-based cytotoxicity assay

Cell viability following exposure to synthetic compounds was estimated by using the MTT reduction assay (19-20). K562 and LS180 cells were plated in 96-well microplates at a density of 5×10^4 cells/mL (100 µL per well), While MCF-7 cells were plated at densities of 3×10^5 . Control wells contained no drugs and blank wells contained only growth medium for background correction. After overnight incubation at 37 °C, half of the growth medium was removed and 50 µL of medium supplemented with 3 different concentrations of synthetic were added in duplicate. Plates with K562 cells were centrifuged before this procedure. Compounds were all first dissolved in DMSO and then diluted in the growth medium. The maximum concentration of DMSO in the wells was 0.5%. Cells were further incubated for 72 h and at the end of the incubation time; the medium was replaced with fresh medium containing 0.5 mg/

mL of MTT. Plates were incubated for another 4 h at 37 °C. Then the formazan crystals formed in the cells dissolved in 200 µL DMSO. The optical density was measured at 570 nm with background correction at 655 nm using a Bio-Rad microplate reader (Model 680). The percentage inhibition of viability for each concentration of compound was calculated compared to the control wells and IC₅₀ values (concentration of the compound that induces 50% inhibition of cell viability) were calculated with the CurveExpert software version 1.34 for Windows. Each experiment was repeated 4 times. Data are presented as mean ± S.D.

Conformational study

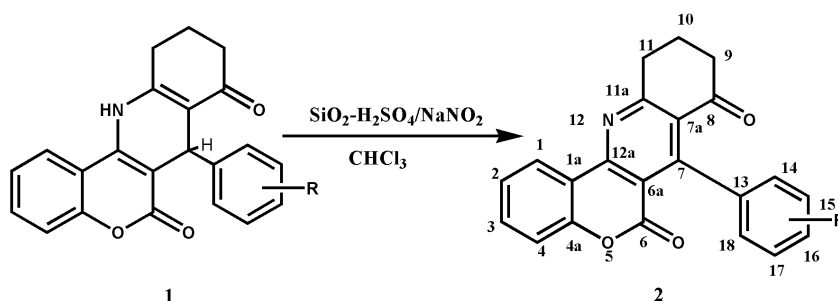
The chemical structure of molecules was constructed using Hyperchem (Version 7, Hypercube Inc., <http://www.hyper.com>, USA). The Z-matrices of the structures were provided by the software and were then transferred to the Gaussian 98 program (21). Complete geometry optimization was performed taking the most extended conformations as starting geometries. Semi empirical molecular orbital calculations (AM₁) of the structures were performed using Gaussian 98 program. Then we calculated some important dihedral angles of this optimized structure such as orientation of C₇-aryl substitute and conformation of cyclohexenone ring.

Results and Discussion

In this paper, we aromatized synthesized dihydropyridines (DHPs) (1) in the presence of oxidizing reagent, silica sulfuric acid/NaNO₂ to corresponding new pyridine derivatives (2), (Scheme 1).

We first used the most common methods of oxidation by means of oxidizing agents such as Mn(IV), Ti(IV) and V(V) for the oxidation of compound 1, but all these methods proved unsuccessful. However, compound 1 could be aromatized to compound 2 using silica sulfuric acid and sodium nitrite in boiling chloroform with 64-74 yields. All reactions were completed in an appropriate time and gave only the corresponding pyridine derivatives. The results are summarized in Table 1.

The proposed mechanism involved two



Scheme 1. synthetic route for oxidative aromatization of chromeno[4, 3-b]quinoline.

steps; nitrosation of DHP by nitrous acid and aromatization by losing hydrogen and NO (Scheme 2). Nitrosation reaction was performed under mild and heterogeneous conditions. In this reaction, wet SiO_2 acts as a media and provides a heterogeneous effective surface area for in situ generation of HNO_2 .

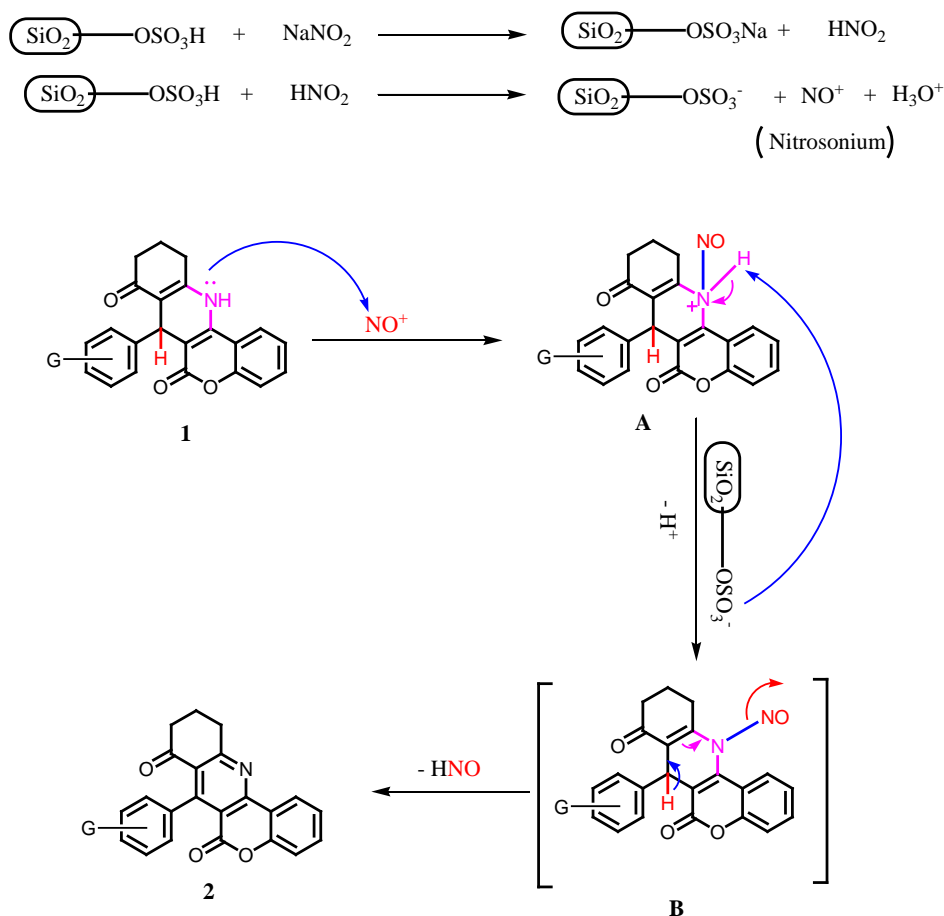
The structures of compounds were confirmed by ^1H NMR, ^{13}C NMR and EI-MS spectra. The ^1H NMR spectra showed the high deshielding character of aliphatic protons from 1.8-2.9 in 1 to 2.2-3.5 in 2 confirming the formation of pyridine. The protons belonging to CH and NH in dihydropyridine of compounds 1 was disappeared in compounds 2. Mass spectrum and elemental analysis clearly supports the proposed structure.

The cytotoxic activity of synthesized compounds was evaluated in three different

human cancer cell lines including K562 (chronic myelogenous leukemia), LS180 (colon adenocarcinoma) and MCF-7 (breast adenocarcinoma). Data are demonstrated in Table 2. Compounds showed moderate cytotoxic activities. The most active one appeared to be 2e, containing a methoxy group on the meta position of phenyl ring, with the lowest IC_{50} values (11.1, 26.8 and 55.7 μM on LS180, MCF-7 and K562 cells, respectively). When comparing this most potent compound with the dihydropyridine derivative counterpart of our previous study (14) (1e containing methoxy group on meta position of phenyl ring) an interesting result is achieved; 1e is not active on any of these cell lines (the IC_{50} of 1e is greater than 100 μM in all three cell lines). The other compounds 2c, 2m and 2k also showed good activities. All of these compounds are more potent than the corresponding dihydropyridine

Table 1. Reactions time and yields of oxidation reaction.

Compound	R	Time(h)	Yield (%)	MP(°C)
2a	2- CH_3	8	71	210-212
2b	3- CH_3	8	64	172-174
2c	4- CH_3	8	68	241-243
2d	2- OCH_3	8	71	180-182
2e	3- OCH_3	8	66	155-157
2f	4- OCH_3	8	75	185-187
2g	2-Cl	9	66	253-255
2h	3-Cl	8	68	188-190
2i	4-Cl	8	70	281-283
2j	2- NO_2	12	70	161-163
2k	4- NO_2	12	67	265-267
2l	3-Br	8	64	184-186
2m	4-Br	8	65	279-281
2n	H	8	74	121-123



Scheme 2. Probable Mechanism pathway for the aromatization of Chromeno [4, 3b]quinoline.

Table 2. Cytotoxic activity of synthetic compounds assessed by the MTT reduction assay.

Compound	IC ₅₀ (μM)		
	K562 cells	LS180 cells	MCF-7 cells
2l	137.1 ± 48.1	> 200	120.8 ± 29.8
2n	116.1 ± 31.7	> 200	90.6 ± 23.1
2f	> 200	46.9 ± 9.1	85.4 ± 21.7
2d	> 200	> 200	> 200
2i	67.6 ± 9.4	> 200	55.0 ± 20.5
2e	55.7 ± 26.7	11.1 ± 3.5	26.8 ± 5.9
2c	89.9 ± 43.1	21.9 ± 3.6	63.5 ± 13.9
2h	> 200	> 200	133.6 ± 30.2
2g	> 200	> 200	> 200
2a	132.0 ± 13.5	> 200	> 200
2k	147.5 ± 7.6	38.7 ± 12.1	66.4 ± 14.1
2m	52.5 ± 13.6	41.9 ± 6.5	43.8 ± 6.9
2b	> 200	> 200	> 200
Cisplatin	6.5 ± 0.5	15.5 ± 1.9	15.7 ± 9.6

Values represent the mean ± S.D. of 3 to 4 different experiments.

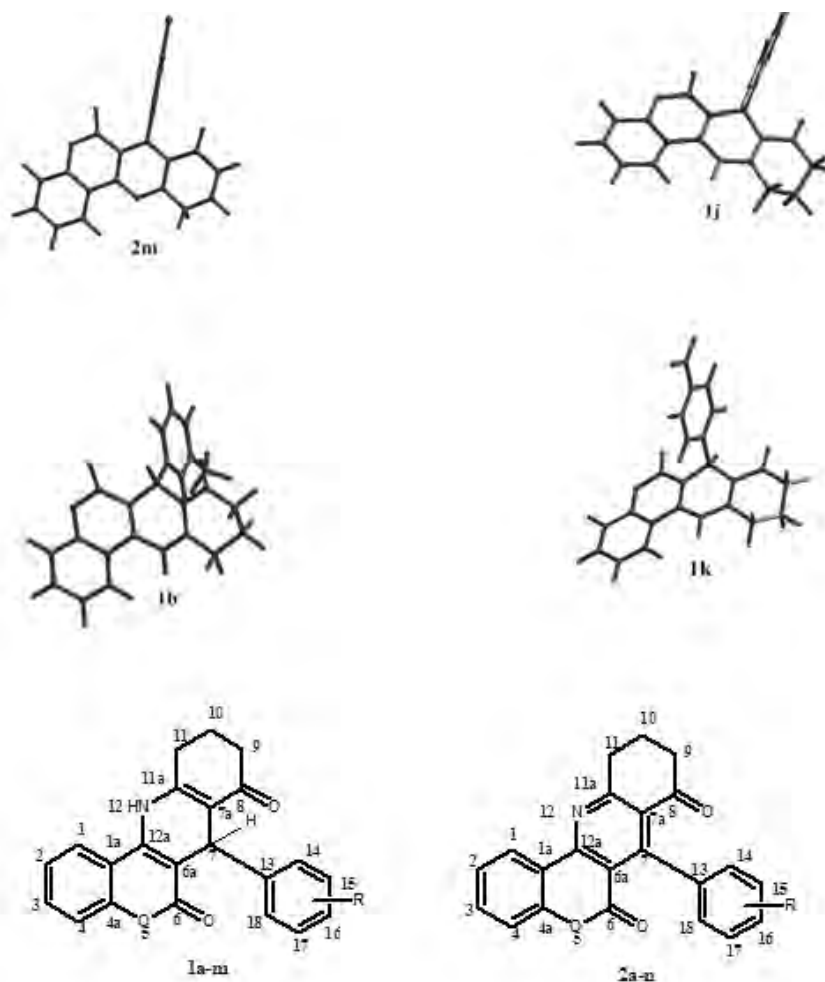


Figure 1. The optimized three-dimensional structural representation of six selected compounds.

derivatives especially in LS-180 cell line (the corresponding IC_{50} of 2c, 2m and 2k in LS-180 cell line is 21.9, 41.9 and 38.7 μM respectively). However, 1c, 1m and 1k were not active in this cell line and 2d, 2g and 2b compounds were inactive on all three cell lines (14).

By comparing the pyridine derivatives with the non-oxidized derivatives of our previous study (14), it can be concluded that the cytotoxic activity of some pyridine derivatives are improved especially in LS180 cell lines. Therefore the condensation and aromatization of the structure resulted in enhanced antitumoral activity in most of studied compounds.

The optimized 3D structures of molecules (1a-n and 2a-m) were obtained by semi-empirical molecular orbital calculations (AM1). Structures

of six compounds are presented in Figure. 1. The calculated dihedral angles are represented in Table 3.

The cyclohexenone ring exhibited a semi-boat conformer in all studied derivatives. The “ $C_{6a}-C_7-C_{13}-C_{14}$ ” dihedral angle reflects the orientation of aryl group at C_7 position. As it can be seen in Table 3, the aryl group positioned at the axial coordinate especially in oxidized derivatives 2a-m (the relevant dihedral angle is 90-100°). However, some deviation from axial orientation is seen in non-oxidized group 1a-n. Besides, the deviation of aryl ring from axial position is clear in less potent compounds. Moreover, it seems that the main difference of pyridine 2a-m and dihydropyridine 1a-n derivatives is the spatial orientation of C_7 -aryl ring with respect to the main structure (Table

Table 3. Geometrical features of the optimized structures of synthesized compounds.

Compound	R	pIC50 K562 cells	pIC50 LS180 cells	pIC50 MCF7 cells	$\Phi 1^a$ ($^\circ$)	$\Phi 2^b$ ($^\circ$)	$\Phi 3^c$ ($^\circ$)
1a	2-CH ₃	NA ^d	NA	ND ^e	35.97	60.69	-102.43
1b	3-CH ₃	4.24	4.28	ND	36.69	134.21	-103.23
1c	4-CH ₃	4.25	4.18	ND	36.90	58.15	-102.18
1d	2-OCH ₃	NA	NA	ND	36.71	-123.64	-102.18
1e	3-OCH ₃	NA	NA	ND	44.75	51.99	-100.60
1f	4-OCH ₃	NA	4.21	ND	35.81	-123.56	-101.88
1g	2-Cl	NA	NA	ND	37.45	55.95	-102.13
1h	3-Cl	4.29	4.14	ND	37.41	-145.51	-104.01
1i	4-Cl	4.43	4.38	ND	38.59	60.22	-103.33
1j	2-NO ₂	4.60	4.23	ND	20.69	-63.87	101.47
1k	4-NO ₂	4.36	NA	ND	15.71	130.28	113.42
1l	3-Br	4.41	4.50	ND	23.11	-67.25	101.24
1m	4-Br	4.32	4.34	ND	20.25	-61.37	101.71
2a	2-CH ₃	3.88	NA	ND	-33.88	-92.55	-179.32
2b	3-CH ₃	NA	NA	ND	-52.81	-75.72	-173.63
2c	4-CH ₃	4.05	4.66	4.20	-47.26	-94.67	-179.92
2d	2-OCH ₃	NA	NA	NA	-53.46	-77.95	-150.79
2e	3-OCH ₃	4.25	4.95	4.57	-40.98	-101.115	-177.93
2f	4-OCH ₃	NA	4.33	4.07	-47.36	-96.79	-179.28
2g	2-Cl	NA	NA	NA	-49.20	-91.61	-173.87
2h	3-Cl	NA	NA	3.87	-39.98	-95.41	-178.63
2i	4-Cl	4.17	NA	4.26	-47.47	-93.20	-176.10
2j	2-NO ₂	NA	NA	NA	-37.88	-96.99	-178.68
2k	4-NO ₂	3.83	4.41	4.18	-41.58	-98.02	-178.94
2l	3-Br	3.86	NA	3.92	-43.05	-93.33	-178.58
2m	4-Br	4.28	4.38	4.36	-50.73	-81.2	-173.87
2n	H	3.94	NA	4.04	-44.02	-94.73	-179.06

^a Dihedral angle of cyclohexenone ring (C_{7a}-C₈-C₉-C₁₀)^b Dihedral angle between phenyl and pyridine (or dihydropyridine) rings (C_{6a}-C₇-C₁₃-C₁₄)^c Dihedral angle between phenyl and pyridine (or dihydropyridine) rings (C_{12a}-C_{6a}-C₇-C₁₃)^d Inactive^e Not determined

3). The dihydropyridine series 1a-n show approximately antiplanar position with respect to four main ring structure (the relevant dihedral angle is 90-100°). While in pyridine series 2a-m, the aryl ring adopt synplanar orientation with respect to main four ring structure in most cases ($\Phi 3$ is near 180° or -180°).

Conclusion

In search of novel antitumoral compounds, a set of 7-aryl-10,11-dihydro-7H-chromeno [4,3-

b] quinoline-6,8(9H, 12H)-dione derivatives were synthesized by a simple one-pot method using silica sulfuric acid/NaNO₂ as an oxidative agent for aromatization of 1,4-DHPs. The cytotoxic activity of these compounds was evaluated *in-vitro* on three different cancer cell lines (K562, LS180, and MCF-7). Most of synthetic compounds showed moderate cytotoxic activities. Comparison of the cytotoxic activity of these novel oxidized derivatives with non-oxidized counterpart revealed that oxidation of dihydropyridine ring to pyridine, improves

the cytotoxic activity especially in LS180 cell line. Conformational analysis revealed that some conformational aspects of oxidized derivatives such as orientation of C₇-aryl were clearly different from non-oxidized ones. Therefore, these novel condensed derivatives seem to have promising anticancer properties and further investigation on this group, especially by optimization of heterocyclic and aromatic rings of the structure, could potentially lead to the discovery of potent cytotoxic agents.

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