



# GC-MS Analysis of Anti-*Candida* and Antioxidant Activities of Hydroalcoholic Leaf Extract of *Chaerophyllum macropodum*

Farahnaz Bineshian<sup>1</sup>, Neda Bakhshandeh<sup>2</sup>, Kazem Taherian<sup>3</sup> and Hossein Nazari<sup>2,\*</sup>

<sup>1</sup>Department of Mycology, Faculty of Medicine, Semnan University of Medical Sciences, Semnan, Iran

<sup>2</sup>Department of Biochemistry, Faculty of Medicine, Semnan University of Medical Sciences, Semnan, Iran

<sup>3</sup>Department of Natural Plants Herbarium, Semnan Agriculture and Natural Resources Research Center, University of Applied Science and Technology, Semnan, Iran

\*Corresponding author: Department of Biochemistry, Faculty of Medicine, Semnan University of Medical Sciences, Semnan, Iran. Tel: +98-2313353170-3; Fax: +98-2313354161; Email: hosnazbio@yahoo.com

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## Abstract

**Background:** *Chaerophyllum* is a genus from the family Apiaceae. *Chaerophyllum macropodum* as a wild species is endemic to Iran and Turkey. The leaves of this plant are used in the food industry, as well as medicine. Since resistance has increased in some *Candida* species, side effects of chemical drugs and the use of medicinal plants are very important.

**Objectives:** We reviewed the antioxidant and anti-*Candida* effects of hydroalcoholic leaf extracts of *C. macropodum*.

**Methods:** For the evaluation of the antioxidant potential of *C. macropodum* extracts, DPPH radical scavenging test, reducing power assay, and  $\beta$ -carotene bleaching test were performed. Moreover, the total free phenolic and flavonoid contents were measured. A GC-MS instrument was used for analyzing the extract components. In addition, hydroalcoholic extraction was performed by agar well-diffusion method against 40 *Candida* species isolated from patients.

**Results:** According to the DPPH test, a moderate antiradical activity was attributed to *C. macropodum* leaf extract; maximum inhibition (6.38) was reported at 1000  $\mu\text{g}/\text{mL}$ . Based on the findings, the total phenolic and flavonoid contents were  $57.43 \pm 0.25$  and  $138 \pm 19.80$  (mg/g), respectively. The minimum inhibitory concentrations for *C. albicans* and *C. glabrata* were 300 and 400  $\mu\text{g}/\mu\text{L}$  at 48 hours, respectively.

**Conclusions:** In this study, a moderate antioxidant activity was attributed to *C. macropodum* leaf extracts. In addition, the results confirmed that leaf extracts had inhibitory activity against bacteria and *Candida* species. Therefore, they may be applied as antioxidant, antibacterial, and antifungal agents in food protection.

**Keywords:** *Chaerophyllum*, *Candida*, Antioxidant Activity, Hydroalcoholic Extract

## 1. Background

*Chaerophyllum* genus, from the Apiaceae family, includes nearly 110 species in temperate and sub-temperate regions of Asia, Africa, and Europe. *Chaerophyllum* (chervil) is a native herb with weak stems and leaves, similar to parsley or parsley-like leaves. *C. macropodum* is known as "Gazarsare" in Mahdishahr (Sangsar) and "Jafari Frangi Kohi" in Iran. Eight *Chaerophyllum* species have been found in Iran, including *C. macropodum*, which is a wild species endemic to Iran and Turkey. The leaves of this plant are used in the food industry, as well as medicine. Previous studies have confirmed the presence of phenolic acids and constituents with cytotoxic and antimicrobial activities in *Chaerophyllum* species (1-4).

*Candida* species, particularly *Candida albicans*, are common pathogens in patients with organ transplantation, AIDS, and neoplastic disease, immunocompromised pa-

tients, and users of immunosuppressive drugs and broad-spectrum antibiotics. Recently, the use of medicinal plants has been highlighted, as natural products with therapeutic properties are cost-effective, available, and beneficial. In addition, chemical drugs have more side effects. Treatment of some diseases using medicinal plants, especially herbs, is in its infancy. On the other hand, researchers try to detect the active constituents of extracts to treat different diseases.

## 2. Objectives

We aimed to examine the anti-*Candida* and antioxidant potential of hydroalcoholic leaf extracts of *C. macropodum* via GC analysis.

### 3. Methods

#### 3.1. Materials

Sigma-Aldrich supplied Gallic acid, trichloroacetic acid (TCA), vitamin C, Quercetin, DPPH, Folin-Ciocalteu reagent, and butylated hydroxytoluene (BHT). The spectrophotometric assays were performed with a spectrophotometer (Analytik Jena, Germany).

#### 3.2. Collection of Plants

After collecting *C. macropodum* leaves from Roudbarak in Mahdishahr (Semnan, Iran), they were confirmed by professional botanists. A voucher sample was deposited at the Research Center for Medicinal Plants. The dried samples were powdered for further analysis.

#### 3.3. Extract Preparation

For extraction, the powdered leaf (50 g) was mixed with 6:3:1 ethyl acetate: methanol: distilled water (500 mL) overnight. The temperature was adjusted to 40°C in a Soxhlet system for eight hours before the evaporation solvent was incubated at room temperature.

#### 3.4. Total Phenolic Content (TPC)

The Folin-Ciocalteu method was used according to the literature (5, 6) with some modifications to evaluate TPC. Following two hours of incubation, absorbance was read at 765 nm. For the measurements, the standard Gallic acid (GA) curve was applied, presented as grams of GA equivalent (GAE) per 100 g of dry weight (7).

#### 3.5. Total Flavonoid Content (TFC)

To determine TFC, the extract (1 mL) in methanol was mixed in dilutions of standard quercetin solution in a flask (10 mL) of distilled water (4 mL). During five minutes of incubation, 5% sodium nitrite (0.3 mL) was added. In addition, aluminum chloride (0.3 mL) was added before six minutes of incubation, followed by 1 M sodium hydroxide (2 mL). OD was determined at 510 nm against a blank reagent. TFC was expressed as mg of quercetin equivalent per gram of sample (8-10). To obtain the extract concentrations, the standard curve of quercetin was used. TFC was calculated as follows:

$$TFC = R \times D.F \times V/w$$

where R represents the standard curve result, D.F denotes the dilution factor, V indicates the stock solution volume, and W represents the plant weight.

#### 3.6. Antioxidant Assay

For the antioxidant assay, after dissolving the sample in 95% methanol at 1 mg/mL, it was diluted for preparation of series concentrations. The reference chemicals were used for comparisons.

##### 3.6.1. DPPH Radical Scavenging Assay (DRSA)

DRSA was conducted and the absorbance was read at 517 nm based on studies by Burits and Bucar (11), Cuendet et al. (12), and Kirby and Schmidt (13):

$$\text{DPPH radical scavenging percentage: } \% = (1 - A_s/A_c) \times 100$$

where  $A_c$  indicates the control absorbance, while  $A_s$  denotes the sample solution absorbance.

##### 3.6.2. Reducing Power Assay

For this purpose, the method proposed by Hinneburg et al. (14) was applied. To measure absorbance, a spectrophotometer was used at 700 nm after mixing the supernatant (2.5 mL) with  $\text{FeCl}_3$  (0.1%, 0.5 mL) and distilled water (2.5 mL). An increase in absorbance was indicative of increased reducing activity.

##### 3.6.3. $\beta$ -Carotene Bleaching Test (BCBT)

To prepare the stock solution of linoleic acid and  $\beta$ -carotene, linoleic acid (25  $\mu\text{L}$ ), as well as Tween 40 (200 mg), was added after dissolving  $\beta$ -carotene (0.5 mg) in chloroform (1 mL), since  $\beta$ -carotene is not considered water-soluble. Following that, 100 mL of oxygen-saturated distilled water was added during 30 minutes via vigorous shaking at 100 ml/min. Finally, the mixture (2500  $\mu\text{L}$ ) was added to tubes and then, the extract (350  $\mu\text{L}$ ) was added at 2 g/L.

OD was measured at 490 nm after incubation. A similar procedure was applied using a positive BHT control, along with a blank. Comparisons were made in terms of antioxidant potential between the extract, BHT, and the blank. All assays were conducted three times, and the relative antioxidant activity (RAA) was estimated as follows (15):

$$RAA = A_{\text{sample}}/A_{\text{BHT}}$$

#### 3.7. Preparation of Microbial Strains and Inoculum

##### 3.7.1. Serial Dilutions

For a final concentration of 400  $\mu\text{g}/\mu\text{L}$ , 2 g of *C. macropodum* extract was dissolved in 5 mL of DMSO (16). To prepare serial dilutions of 400, 350, 300, 250, 200, 150, and 100  $\mu\text{g}/\mu\text{L}$ , sterile distilled water was used. The reference strain was *Candida albicans* (ATCC10231). Gram-positive and Gram-negative bacteria included *Staphylococcus aureus* and *Escherichia coli*, respectively. Finally, 40 *Candida*

isolates were obtained from women with vulvovaginal candidiasis, including 34 *C. albicans* and six *C. glabrata* isolates (17).

After culturing the bacterial strains in nutrient agar (NA), incubation was performed for 24 hours at 37°C. Following the inoculation of the *Candida* isolates in Sabouraud dextrose broth (SDB, Merck, Germany), incubation was performed for 24 hours at 35°C. Then, using sterile distilled water, the yeasts were washed three times and the suspension was used to prepare the  $1 \times 10^6$  CFU/mL concentration (0.5 MF standard). The positive and negative controls included the fluconazole disc for fungi (10 µg/disc; Hi-Media, India) and gentamicin (10 µg/disc) for bacteria, and DMSO, respectively.

### 3.7.2. Broth Microdilution for MIC Measurements

After adding the yeast inoculum (100 µL,  $10^6$  cells/mL) on SDA (Merck, Germany) and bacterial inoculum (100 µL,  $10^6$  cells/mL) on Mueller-Hinton agar (MHA, Merck, Germany), a borer was used to punch six 7-mm wells in the SDA medium; then, 100 µL of the extract dilutions was added, along with DMSO 100% as the negative control (16). The zone of inhibition was measured after incubating the plates at 35°C for 24 hours to determine the anti-*Candida* activity. The experiments were repeated twice.

### 3.8. GC-MS Analysis

A GC-MS system (Agilent) with a VF-5 ms silica capillary column (DB-35 ms, 30 m × 250 µm × 0.15 µm) was used for the GC-MS analysis (18), using helium gas (99.99%) as the carrier at a constant flow of 1 mL/min; the temperature of the injector and transfer line was 200°C and 250°C, respectively. On the other hand, the oven temperature increased from 60°C to 280°C at 10°C/min. After maintaining the isothermal temperature for 60 seconds, it increased to 280°C at 10°C/min. In the next step, the sample (1 µL) was injected in the splitless mode (50 - 600 amu; split ratio, 1:10). The total running time was 57 minutes. The retention indices, as well as fragmentation patterns, were compared with the computer library data and literature to identify the extract components. Using the Willey library, the recognized constituents were matched.

### 3.9. Statistical Analysis

One-way ANOVA was applied at 95% confidence level in SPSS version 16. In addition, the correlation of antioxidant activity with TPC and TFC was determined via linear regression analysis in Excel 2003.

## 4. Results

### 4.1. Extraction Yield, TPC, and TFC

The extraction method, growth, and climatic conditions, as well as genetic factors, influence TPC, TFC, and antiradical activity. However, according to the literature (18), the use of a different solvent system, i.e., ethyl acetate: methanol: distilled water, is preferable for the extraction of antioxidant constituents. The percentage of hydroalcoholic extraction yield of *C. macropodium* leaf extract was 1.86 w/w%.

### 4.2. Antioxidant Potential

The high antioxidant potential was confirmed on DRSA, reducing power assay, and BCBT.

#### 4.2.1. DRSA

In this primary assay, odd electrons of DPPH are quenched by the extract to decrease absorbance at 517 nm. The leaf extract of *C. macropodium* showed high antiradical activity in comparison with the standard BHT with the highest inhibition at 1000 µg/mL (14.47) (Table 1).

#### 4.2.2. Reducing Power Activity

Compared to vitamin C, the reducing potential of the hydroalcoholic extract was not dose-dependent. The reducing power percentage is demonstrated in Table 1. The leaf extract of *C. macropodium* showed maximum reducing power ( $P \leq 0.05$ ). Analysis of the association between the reducing power activity and phenolic content indicated an  $R^2$  of 0.89, while its correlation with the flavonoid content was estimated at  $R^2 = 0.88$ . Based on these values, phenolic compounds may have greater involvement in the reducing activity, compared to flavonoid compounds.

#### 4.2.3. BCBT

The intensity of yellow color will reduce following the reaction between β-carotene and radicals produced via linoleic acid oxidation. The RAA of *C. macropodium* extract was measured as follows:

$$RAA = A_{\text{sample}}/A_{\text{BHT}}$$

where  $A_{\text{BHT}}$  represents the BHT absorbance (standard) and  $A_{\text{sample}}$  denotes the absorbance of *C. macropodium*. The calculated RAAs for *C. macropodium* leaf extract and BHT are presented in Table 1.

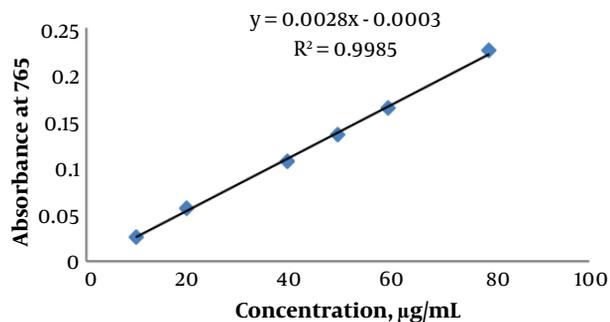
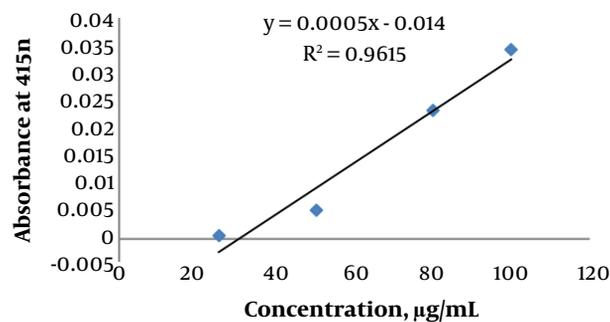
### 4.3. Bioactive Content

#### 4.3.1. TPC Measurements

For the spectrophotometric measurement of TPC (expressed as GAE), the Folin-Ciocalteu method was applied

**Table 1.** Effect of the Test Compounds in the DPPH Assay,  $\beta$ -Carotene Bleaching, and Reducing Power of the Hydroalcoholic *C. macropodum* Extract with Respect to Standard Vit.c and BHT

Test Compound	IC <sub>50</sub>	Inhibition (%)	$\beta$ -Carotene Bleaching (RAA%)	Reducing Power (%)
<i>C. macropodum</i>	27.63	6.38	37.61 $\pm$ 0.92	94.48 $\pm$ 3.75
BHT	14.55	14.47	100	

**Figure 1.** Calibration curve of standard Gallic acid for determination of total phenol content**Figure 2.** Calibration curve of standard Quercetin for determination of total flavonoid content in *C. macropodum*

and the standard curve equation was determined (Figure 1):

$$\text{Absorbance} = 0.038 \times (\mu\text{g GA}) - 0.0132 \quad (R^2 = 0.9859)$$

TPC was calculated by adding the absorbance value in this equation. As shown in Table 2, TPC of *C. macropodum* leaf extract was  $57.43 \pm 0.25$ .

**Table 2.** Antioxidant Constituents of *Chaerophyllum macropodum*<sup>a</sup>

Extract	Total Phenols (GAE) (mg/g) <sup>b</sup>	Total Flavonoids (QU) (mg/g) <sup>c</sup>
<i>C. macropodum</i>	57.43 $\pm$ 0.25	138 $\pm$ 19.80

<sup>a</sup> Values represent means  $\pm$  standard deviations for triplicate experiments.

<sup>b</sup> Expressed as GAE.

<sup>c</sup> Expressed as Quercetin equivalent (QU).

#### 4.3.2. TFC Measurements

For the spectrophotometric measurement of TFC, the Folin-Ciocalteu method was applied and the standard curve equation was measured (Figure 2):

$$\text{Absorbance} = 0.0005 \times (\mu\text{g quercetin}) - 0.014 \quad (R^2 = 0.9615)$$

TFC was calculated after entering the absorbance value in the equation. The mean TFC was  $138 \pm 19.80$  mg/g (Table 2).

#### 4.4. Anti-Candida Activity

The anti-*C. albicans* (n = 34) and anti-*C. glabrata* (n = 6) activities of hydroalcoholic extracts were confirmed in our analyses (Table 3).

#### 4.5. GC-MS Analysis

The presence of phytochemicals was confirmed based on the GC-MS analysis. 15 compounds were detected in *C. macropodum* (Table 4).

### 5. Discussion

Recently, the use of medicinal plants has attracted the researchers' attention. Polyphenols are secondary metabolites, scavenging and quenching free radicals. Coruh et al. (4) showed that TPC of *C. macropodum* was 34  $\mu\text{g GA/mg}$ . Ebrahimabadi et al. (19) reported that TPC was 29.3 and 30.2  $\mu\text{g GA/mg}$  in the methanolic leaf and flower extracts of *C. macropodum*, respectively. According to a study by Alavi et al. (20), TPC of the dried extract of *C. macropodum* was 1.34  $\mu\text{g/mg}$ . In our study, TPC was 57.43  $\mu\text{g GA/mg}$  in the hydroalcoholic leaf extracts of *C. macropodum*, which is higher than the level reported in the mentioned studies.

On the other hand, 15 compounds were identified in the hydroalcoholic leaf extracts of *C. macropodum* in this research. The major chemical constituents included palmitic acid, linolenic acid, linoleic acid, nonacosane, trimethyloctane, ethyl linoleate, neophytadiene, benzothiazole, pyridinecarboxaldehyde, and oleic acid. Based on the findings, omega-3, -6, and -9 and palmitic acid were respectively unsaturated and saturated fatty acids. Unsaturated fatty acids showed higher concentrations than saturated fatty acids. The GC-MS analysis showed that

**Table 3.** Antimicrobial Activity of the Hydroalcoholic Extract of *C. macropodum*

Variable	Inhibition Zone	MIC	Gentamicin	Fluconazole
<i>Candida albicans</i>	9	300 µg/µL	-	20
<i>Candida glabrata</i>	8	400 µg/µL	-	-
<i>Escherichia coli</i>	7	400 µg/µL	20	-
<i>Staphylococcus aureus</i>	10	300 µg/µL	21	-

**Table 4.** Chemical Analysis of the Constituents of the Hydroalcoholic Extract of *C. macropodum* Leaves Based on GC Peak Areas

No	Components	Peak Area, %	RT
1	Pentenylamine	0.058	5.308
2	Butanoic acid (active valeric acid)	0.070	8.704
3	Silanol (formic acid)	0.160	11.216
4	4-vinyl-2-methoxy-phenol	0.129	22.162
5	Thiourea	0.108	28.715
6	Neophytadiene	6.524	29.205
7	Hexadecanoic acid (palmitic acid)	27.527	35.866
8	Octadecatrienoic acid (linolenic acid)	20.296	36.175
9	Ethyl linoleate	6.551	36.245
10	Octadecadienoic acid linoleic acid	12.299	39.378
11	Trimethyloctanal	6.588	42.662
12	Pyridinecarboxaldehyde	4.102	44.343
13	Benzothiazole	4.333	44.381
14	9-Octadecenoic acid (oleic acid)	3.522	45.765
15	Nonacosane	7.728	47.165
<b>Total</b>		99.99	

Abbreviation: RT, retention time.

fatty acids were dominant, which is in agreement with a study by Shafaghat (21). Different results regarding the amount of various fatty acids and chemical components may be related to various factors, including season, period of growth, geographical location, altitude, and methods.

Youssef and Noaman (22) showed which benzothiazole derivatives have antibacterial and antifungal activities. Keri et al. (23) conducted a comprehensive systematic review of benzothiazole-based constituents and examined antifungal, antibacterial, antitubercular, antileishmanial, antimalarial, and other medicinal agents. The researchers showed that neophytadiene, a diterpenoid hydrocarbon that belongs to the group of phytanes, has antioxidant and antimicrobial activities (24-26). In the methanolic extract of *Apamarga kshara*, neophytadiene was a superior TvCK inhibitor compared to other identified compounds (27).

Tao et al. (28) showed that octanal derivatives exhibited

strong antifungal activity against *Penicillium* species. In addition, anti-*Candida* and antibacterial activities of some *Chaerophyllum* species showed moderate inhibitory effects (28-30). Ebrahimisadr et al. (31) reported that the extracts of *C. macropodum* have high antileishmania activities. Our results showed that hydroalcoholic leaf extract of *C. macropodum* (nearly 17.3% of the total chemical compound) has an inhibitory activity against bacteria and *Candida*, which is in agreement with the mentioned research and a study by Durmaz et al. (3). It may be due to the presence of compounds, such as benzothiazole, neophytadiene, and octanal, isolated from the hydroalcoholic extract. Therefore, *C. macropodum* is a candidate for the treatment of some diseases and is used in food industries, as well.

On the other hand, the TPC results showed about 57.43 µg GA/mg in hydroalcoholic *C. macropodum* leaf extract, which is higher than the amount reported in other studies (4, 19, 20). Differences in the chemical compounds, content, and therapeutic outcomes related to antifungal and antibacterial activities may be due to various factors, such as season, extraction method, growth period, geographic location, and altitude. There may be a new *Chaerophyllum* species in our region (Rodbarak, Semnan, Iran), based on the plant ecotype theories. Therefore, we will conduct further research on DNA barcoding.

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