



# Evaluation and Comparison of Fullerene (C60) Aqueous Suspension Administration Effects with Memantine HCL in Rat Model of Alzheimer's Disease Considering Behavioral Patterns and Spatial Memory

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Received 2024 January 13; Revised 2024 March 04; Accepted 2024 March 06.

## Abstract

Alzheimer's disease (AD) is among the most prevalent neurodegenerative disorders leading to dementia in the elderly. The accumulation of amyloid-beta ( $A\beta$ ) plaques and the formation of tau protein tangles are primary contributors to AD, which induce oxidative stress. Fullerene C60, a nanoscale carbon allotrope with a diameter of 0.7 nanometers, stands out due to its structure rich in double bonds, making these nanoparticles effective radical scavengers. This property nominates them for potential use in treating neurodegenerative diseases like AD. In this study, unmodified pristine fullerene (C60), a highly hydrophobic molecule, was dispersed in water and administered intraperitoneally (1 mL, BID) to rats after inducing an AD-like condition with scopolamine hydrobromide (2mg/kg, i.p.). The aim was to assess the impact of fullerene (C60) treatment on cognitive behavior and spatial memory in rats, compared to the standard treatment with memantine HCL, using the Morris water maze method. The fullerene aqueous suspension (FAS) was prepared using a solvent exchange method involving a toluene/water mixture and ultrasonication. The concentration of fullerene particles in water was determined by high-performance liquid chromatography (HPLC) to be 21  $\mu\text{g/mL}$ . The Dynamic Light Scattering (DLS) technique measured the average size and zeta potential of the nanoparticles as  $119.14 \pm 3.38$  nm and  $-12.22 \pm 5.98$  mV, respectively. Treatment with FAS significantly improved memory impairment in rats compared to memantine HCL (10 mg/kg, i.p.) treatment. All rats survived until the end of the study, indicating no acute toxicity from FAS administration. These results may offer new insights into combating AD by introducing fullerene C60 as a promising nanoparticle with beneficial effects on behavioral patterns.

**Keywords:** Fullerene, Alzheimer's Disease, Memantine, Scopolamine

## 1. Background

Alzheimer's disease (AD) is one of the most prevalent neurodegenerative diseases leading to dementia in the elderly (1). The loss of neurons results in cognitive disorders, behavioral disturbances, memory dysfunctions, and, at advanced stages, physical disabilities and death (2). The formation of amyloid  $\beta$  ( $A\beta$ ) plaque aggregates and tau protein tangles are primary causes of AD (1), promoting oxidative stress, a leading cause of neuronal toxicity and

death (3). Another critical aspect of AD pathogenesis is the overactivation of glutamate N-methyl-D-aspartate (NMDA) receptors, leading to memory loss and learning difficulties (4). Therefore, inhibiting these receptors could be beneficial in alleviating AD symptoms. Memantine hydrochloride, an uncompetitive antagonist of NMDA receptors, has proven effective in slowing AD progression and improving symptoms through its anti-glutamatergic properties (5). Given the multifactorial nature of AD,

multifunctional therapeutic approaches that enhance the patient's quality of life are highly advantageous (6). In recent decades, healthcare scientists have garnered significant interest in nanotechnology in drug delivery and medicine (7). Fullerene C60, a carbon allotrope with a nanoscale diameter (0.7 nanometers) and a hollow spherical shape, possesses unique physicochemical properties (8). Thanks to its structure rich in double bonds, this nanoparticle is an efficient radical scavenger, making it a candidate for treating neurodegenerative disorders like AD (9). Some studies have suggested that water-soluble formulations of fullerene can prevent the aggregation of A $\beta$  plaques (3) and thereby inhibit the cell apoptosis process that follows plaque aggregate formation (10). Additionally, it has been proposed that fullerenes can reduce the activity of NMDA receptors, although they are not used as NMDA antagonists (9).

Given these characteristics, administering an aqueous suspension of fullerene (C60) in a rat model of AD is anticipated to be highly beneficial for evaluating whether its multifaceted activity can alleviate the disease's symptoms.

## 2. Methods

### 2.1. Chemicals

Fullerene (C60) 99.5% and anhydrous toluene (99.8%) were acquired from Merck (Germany). Acetonitrile (HPLC grade) was sourced from Scharlau (Spain). Memantine hydrochloride was graciously provided by Sobhan Darou Co. (Iran). Wistar rats were procured from the Pasteur Institute (Iran). Scopolamine hydrobromide trihydrate ( $\geq 98\%$ ) was purchased from Sigma-Aldrich (India).

### 2.2. Nano Formulation Preparation

In line with prior studies, a fullerene aqueous suspension (FAS) was prepared (11, 12). Briefly, fullerene (C60) was dissolved in anhydrous toluene at a concentration of 3 mg/mL, producing a dark purple solution. This solution was then mixed with deionized water in a 3:1 ratio. The mixture was stirred for 24 hours at room temperature in a dark setting, after which the toluene was fully evaporated using ultrasonic treatment for several hours under a fume hood. This process resulted in the disappearance of the purple color, leaving behind a pale yellow solution. The final formulation was stored at temperatures between 2 and 8°C in the dark.

To ensure the absence of toluene residue in the formulation, a GC-MS test was conducted, revealing no toluene peaks (Agilent Technologies, 5975C). The size distribution and zeta potential of C60 nanoparticles

dispersed in water were determined at 25°C using the dynamic light scattering (DLS) technique (Brookhaven Instruments). The concentration of C60 nanoparticles in water was measured using high-performance liquid chromatography (HPLC) under the following conditions: Isocratic separation was achieved using a Knauer Smartline 2500 HPLC system with a Eurosphere II -C18 column (250  $\times$  4.6 mm, 100 - 5). The mobile phase consisted of toluene and acetonitrile (60:40, v/v), with a flow rate of 1 mL/min and ultraviolet detection at 328 nm.

### 2.3. Stability Test

A stability test was conducted over 21 days, with assessments every 7 days, to monitor appearance properties such as color and clarity, as well as particle size and zeta potential of the nanoparticles. During each assessment period, zeta potential measurements were taken in duplicate, and size measurements were taken in triplicate for each sample (Figure 1).

### 2.4. Animals

A total of 24 male Wistar rats, weighing between 200 and 240 g, were randomly divided into four groups:

(1) Control group: Received normal saline (i.p.) for 10 consecutive days;

(2) Scopolamine-treated group: Given scopolamine HBr (2 mg/kg, i.p.) for 10 days;

(3) Memantine-treated group: Administered scopolamine HBr (2 mg/kg, i.p.) followed by memantine HCl (10 mg/kg, i.p.) for 7 days;

(4) Fullerene-treated group: Received scopolamine HBr (2 mg/kg, i.p.) for 10 days, followed by a colloidal solution of fullerene C60 (21  $\mu$ g/mL, 1 mL, BID) via i.p. injection for 7 days. Training was conducted on days 18, 19, and 20, and on the 21st day, the rats were tested to determine if they could locate the platform.

The rats were housed under standard cage and laboratory conditions (i.e., temperature at  $23 \pm 2^\circ\text{C}$ , humidity at  $50 \pm 10\%$ , and a 12-hour light-dark cycle), with access to water and food.

### 2.5. Behavioral Learning

To conduct the Morris water maze test, the tank was divided into four equal quadrants (Q1-Q4), with a platform placed in the second quadrant and the tank filled with water.

For each test, the software randomly selected a quadrant of the tank in which to place each rat. Scopolamine hydrobromide, a nonselective muscarinic antagonist, can simulate AD symptoms (13). To induce memory impairment, scopolamine was administered (2

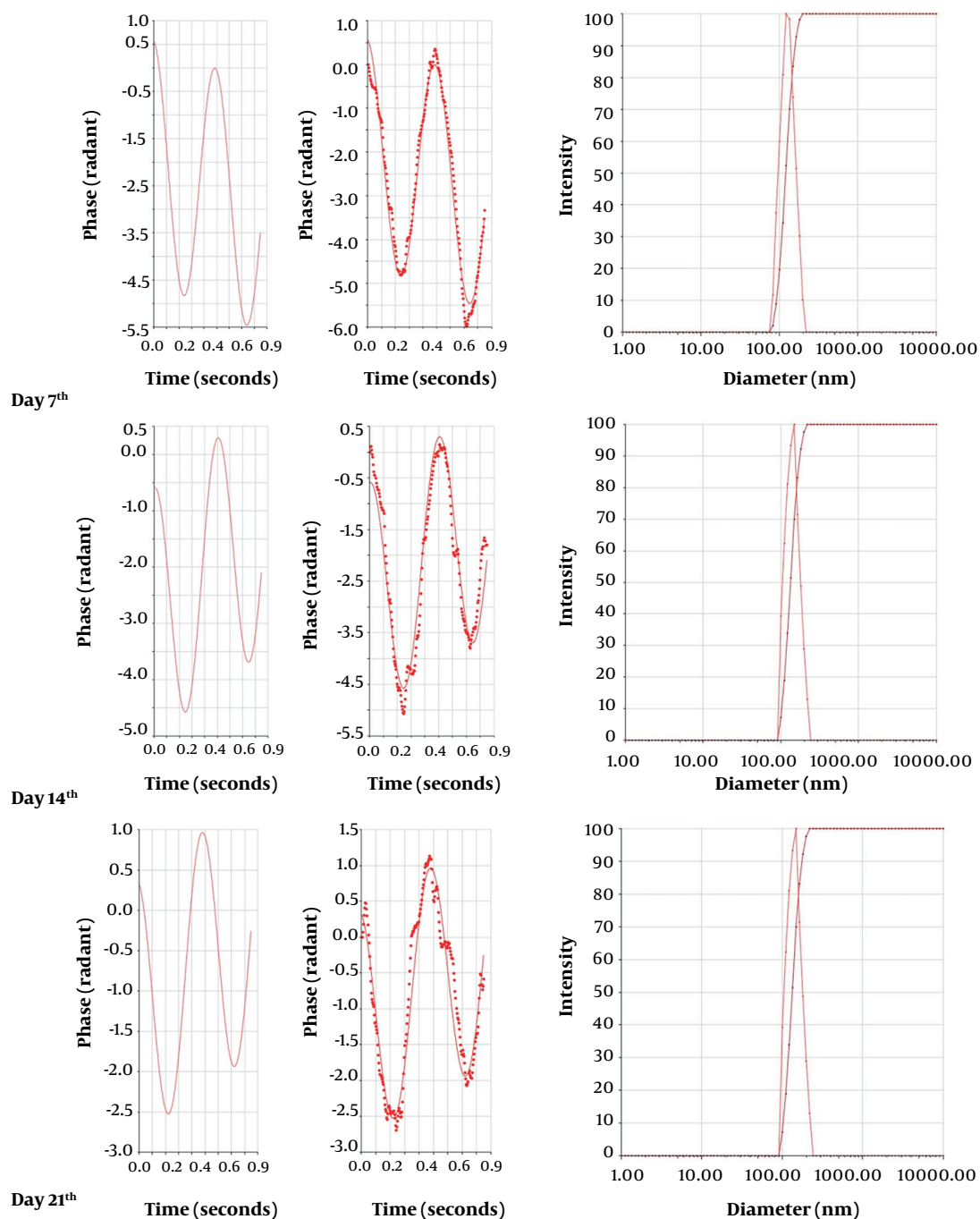


Figure 1. Average particle size and zeta potential variation of fullerene nanoparticles (FAS) over a period of 21 days of stability study.

mg/kg, i.p.) for 10 days to each rat. From the seventh to the ninth day after injection, the rats were placed in the MWM tank for training with a platform in the Q2 segment. On the 10th day, the platform was removed to assess if the rats

could recall the platform's location and to measure the time they spent in the Q2 quadrant.

Their improvement in spatial memory was evaluated based on this parameter (3). Figure 2 shows the

comparison of the test groups, and [Figure 3](#) displays the behavioral pattern graphs of the rats.

### 2.6. Statistics

SPSS 16 was used for data analysis. One-way analysis of variance (ANOVA) and post hoc Tukey tests were conducted to compare groups and interpret the results. A P-value of less than 0.05 was considered significant.

## 3. Results

### 3.1. DLS Data Analysis

The results of the stability test are presented in [Table 1](#). According to the figure, the average particle size, polydispersity index (PDI), and zeta potential were  $119.1 \pm 3.4$  nm, 0.15, and  $-12.22 \pm 6$  mV, respectively.

### 3.2. Fullerene Suspension Calibration Curve

The calibration curve for the fullerene colloidal aqueous suspension was established following the HPLC procedure described earlier. Fullerene standard solutions ranged from 12.5 to 50  $\mu\text{g/mL}$  in a toluene/acetonitrile mixture (60:40 V/V), producing a linear graph ( $Y = 93676.818X$ ;  $n = 5$ ;  $R^2 = 0.9985$ ) with a retention time of approximately 5 minutes ([Figure 4](#)). The limit of detection (LOD) and limit of quantification (LOQ) were 0.39  $\mu\text{g/mL}$  and 1.3  $\mu\text{g/mL}$ , respectively ([Table 2](#)).

### 3.3. HPLC Method Validation

Fullerene standard solutions of 12.5, 30, and 50  $\mu\text{g/mL}$  were prepared in a toluene and acetonitrile mixture (60:40) in triplicates and analyzed in three separate runs. The coefficient of variation (CV%) was calculated for each experiment. The intraday and interday precision (relative standard deviation, RSD%) and accuracy of the HPLC method were determined using calibration curve standards.

### 3.4. Scopolamine Effect of Spatial Memory Suppression

Scopolamine hydrobromide (2 mg/kg) was administered intraperitoneally (i.p.) for 10 days before testing. Spatial memory was assessed by measuring the time rats spent searching for the platform in the Q2 segment. As indicated in [Figures 1](#) and [2](#), the scopolamine hydrobromide-treated group spent less time in the Q2 section compared to the control, positive control, and fullerene-treated groups. This suggests that scopolamine may impair spatial memory.

### 3.5. Memantine HCL Effect on Spatial Memory Improvement

Memantine HCL (10 mg/kg) was administered (i.p.) for a continuous period of 7 days following scopolamine HBr (2 mg/kg) treatment. The effect of the drug on memory impairment was evaluated similarly to the scopolamine group ([Figure 1](#)).

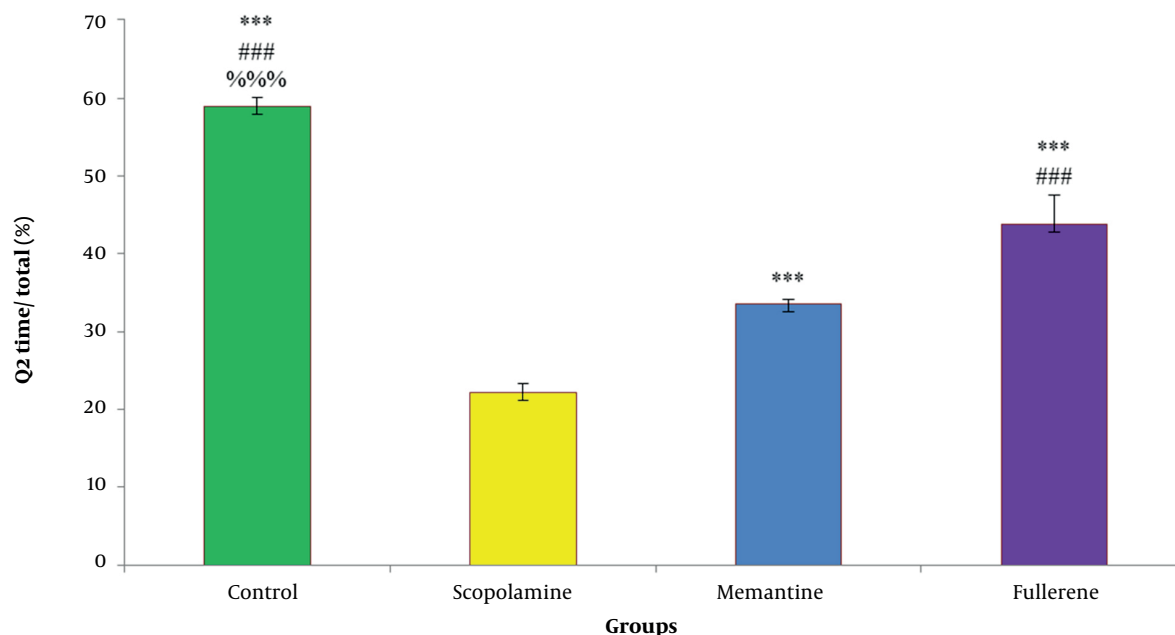
### 3.6. Fullerene Ameliorating Effect on Spatial Memory

After 10 days of scopolamine administration (2 mg/kg), a colloidal solution of fullerene (21  $\mu\text{g/mL}$ , 1 mL, BID) was administered (i.p.) for 7 days. For the following 3 days, rats were trained in the Morris water maze tank, and on the 21st day, they were tested to determine any improvement in memory. According to [Figure 1](#), the fullerene-treated group exhibited significantly better results than the memantine-treated group.

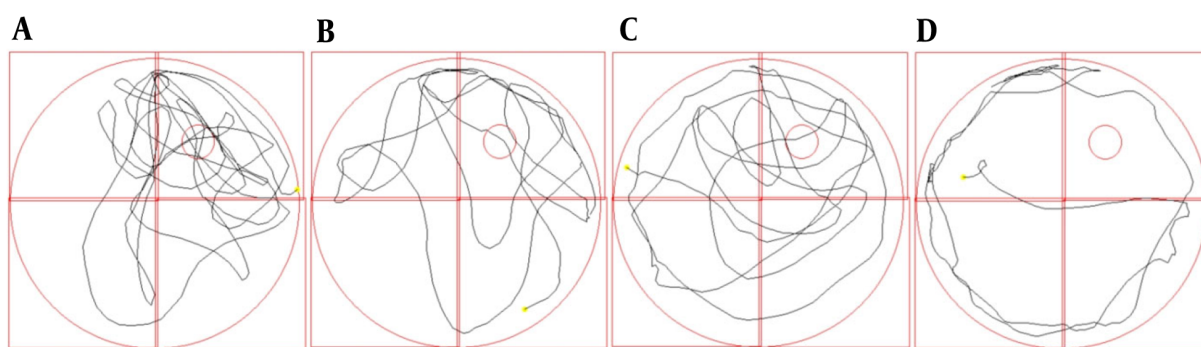
## 4. Discussion

The FAS has been shown to be highly stable from a physicochemical perspective ([14](#)). In our study, the formulation demonstrated satisfactory stability over a 21-day period, as evidenced by the average particle size and zeta potential of the nanoparticles dispersed in water. The zeta potential of the nanoparticles remained relatively unchanged throughout the testing period, whereas there was a significant change in particle size ([Table 1](#)). This alteration could be attributed to the oxidation process of fullerene. Numerous studies have highlighted that water-soluble fullerenes are potent free radical scavengers reacting with reactive oxygen species (ROS). Takada et al., in 2006, conducted extensive research and elucidated this property ([15](#), [16](#)). Therefore, it can be inferred that the increase in zeta potential might result from the addition of positive hydrogen ions to the double bonds of fullerene during the oxidation process, not only making the zeta potential more positive but also leading to an increase in the size of fullerene particles. However, as indicated by previous research, another contributing factor to this phenomenon is the extreme hydrophobicity of fullerenes, which causes them to aggregate and form clusters, in addition to the presence of individual C60 fullerenes ([12](#), [17](#), [18](#)).

Glutamate is a fundamental neurotransmitter in the central nervous system (CNS) ([19](#)). The glutamatergic system, located in the pyramidal neurons of the hippocampus and cortex, plays a crucial role in cognitive processes, memory, and learning by initiating cytosolic calcium signals ([20](#)). In AD, the incidence of pyramidal neurons in the brain is lower than in a normal and healthy condition, indicating a hypoactivity of the



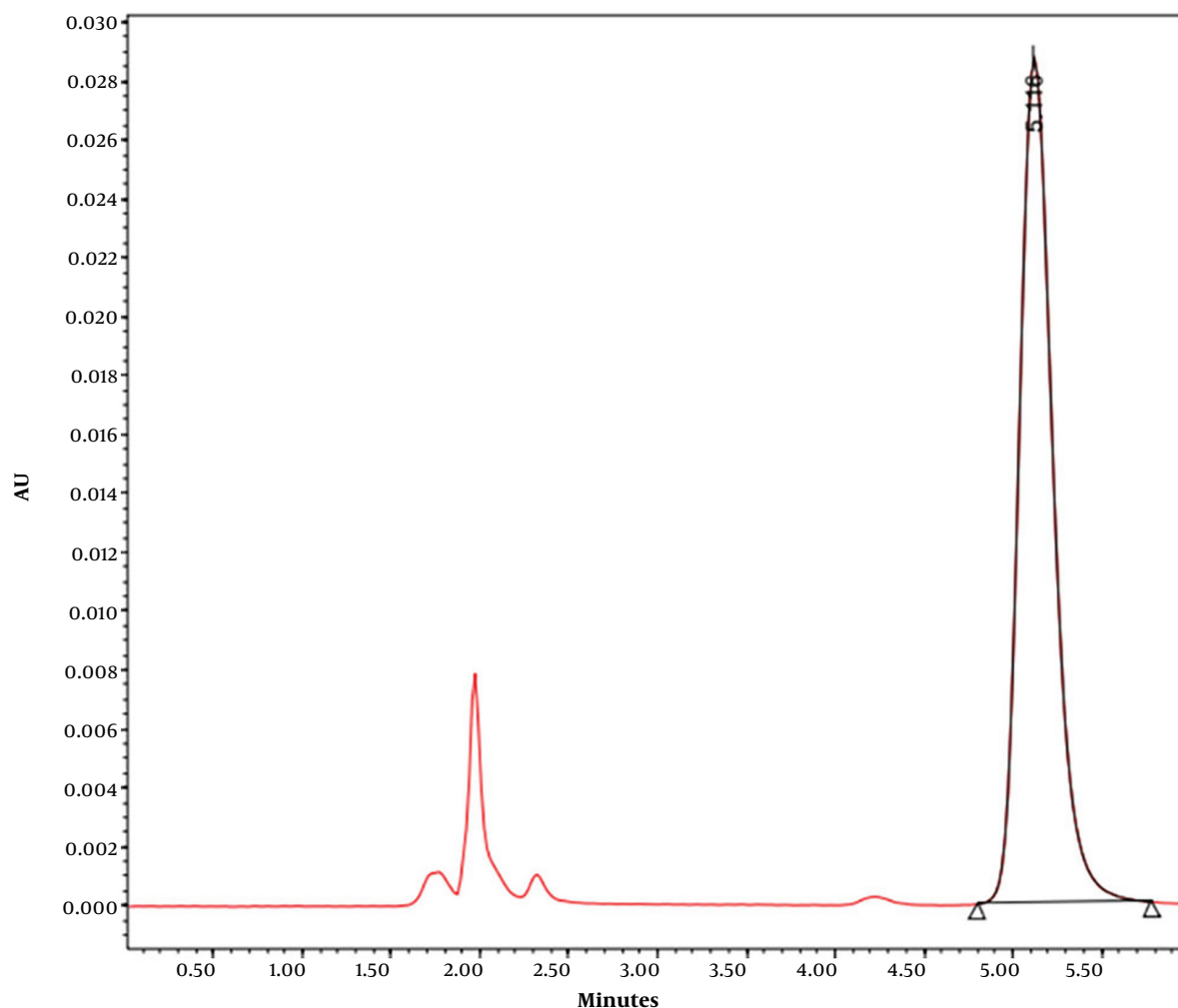
**Figure 2.** Influence of fullerene (C60) administration on spatial memory in rats. Fullerene (21 µg/mL, 1 mL, BID) and memantine HCl (10 mg/kg) were administered (i.p.) for 10 days after scopolamine i.p. injection (2 mg/Kg). Data are represented as mean ± SEM (n = 6 per group). \* P < 0.05 vs. scopolamine, # P < 0.05 vs. memantine, and % P < 0.05 vs. fullerene.



**Figure 3.** Examples of behavioral patterns in different groups of rats treated with (A) Normal saline (1 mL, i.p.), (B) fullerene C60 (21 µg/mL, 1 mL, BID, i.p.), (C) memantine HCl (10 mg/kg, i.p.), and (D) scopolamine hydrobromide (2mg/Kg, i.p.).

**Table 1.** Stability Test of Fullerene (FAS)

Formulation and Day	Color Change	Clarity Change	Average Particle Size, nm	Polydispersity Index	Zeta Potential, mV
<b>Fullerene (FAS)</b>					
1	Yellow	Clear	116.3	0.13	-19.6
7	No	No	116.7	0.16	-13.7
14	No	No	121.6	0.14	-8.3
21	No	No	122.0	0.15	-7.3



**Figure 4.** Chromatogram example of fullerene C60.

**Table 2.** HPLC Method Linearity Condition for Fullerene Analysis

Sample	Concentration Range, $\mu\text{g/mL}$	Slope	Intercept	$R^2$	Concentration Points
Fullerene C60	12.5 - 50	93676.818	-	0.9985	6

glutamatergic system in this disease (19). However, the hyperactivity of glutamate NMDA receptors can lead to several undesirable outcomes, such as severe depolarization of the postsynaptic membrane, calcium overload in mitochondria, dysfunction in mitochondrial operations for expelling oxygen free radicals (ROS) and triggering oxidative stress. All these factors contribute to cell apoptosis and neurodegeneration (21, 22).

Fullerenes and their water-soluble derivatives have demonstrated strong antioxidant activities, the ability to

inhibit neuronal cell apoptosis, and the capacity to block glutamate receptors (23, 24). As potent radical scavengers, they can effectively contribute to the treatment of neurodegenerative diseases (25, 26).

In this study, the effect of water-soluble pristine fullerene (C60) was examined and compared with the effect of memantine hydrochloride on AD. Memantine is known to improve cognitive disorders in AD through the blockage of NMDA glutamate receptors (27). As shown in Figure 1, the administration of fullerene significantly



enhanced cognitive behavior in rats compared to those treated with scopolamine and memantine. This phenomenon is likely related to the antioxidant activity of fullerene as well as its anti-glutamatergic effect (23, 26).

Despite the promising results obtained in this study, a major limitation was the low concentration of fullerene particles dispersed in water. It is anticipated that increasing the yield of the FAS formulation could lead to even better outcomes.

#### 4.1. Conclusions

The findings from this research suggest that the aqueous suspension of fullerene is a highly stable formulation, and its administration can significantly benefit the improvement of cognitive behavior and spatial memory in rats afflicted with AD.

#### Acknowledgments

The authors appreciate the Zanjan University of Medical Science Research Committee that financially supported us through this research.

#### Footnotes

**Authors' Contribution:** Study concept and design: Mehrnoosh Nikpour and Sina Andalib; acquisition of data: Rafi Javadi; analysis and interpretation of data: Hamed Mohammadpour; drafting of the manuscript: Sina Andalib and Mehrnoosh Nikpour; critical revision of the manuscript for important intellectual content: Mehrdad Hamidi; statistical analysis: Hamed Mohammadpour; administrative, technical, and material support: Sina Andalib; study supervision: Mehrnoosh Nikpour and Mehrdad Hamidi.

**Conflict of Interests:** The authors have no conflicts of interest.

**Data Availability:** The dataset presented in the study is available on request from the corresponding author during submission or after publication.

**Ethical Approval:** This study was approved under the ethical approval code [IR.ZUMS.REC.1398.089](#).

**Funding/Support:** This study was financially supported by the Zanjan University of Medical Sciences.

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