

# Isolation of Phenyl Propanoid Glycosides from *Allium tripedale* Trautv

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## ABSTRACT

*Allium* species are widely used plant species, which are consumed worldwide as raw vegetable, to make different dishes and as medicinal plants. As a member of this genus, *Allium tripedale* Trautv., locally called "Khargerui, Anashq", is an important spicy edible *Allium* in "Zagros" region in the west and northwest of Iran which is used widely by local people as a spicy vegetable. Phytochemical investigation of underground part of the plant resulted to the isolation of two phenylpropanoid glycosides as the main phenolic constituents of chloroform-methanol extract of the bulbs. Chemical structure of afforded compounds were elucidated by comprehensive spectroscopic analyses including 1D and 2D NMR and MS as 3-(4-hydroxy phenyl)propyl  $\beta$ -D-Glucopyranoside (**1**) and coniferyl  $\beta$ -D-Glucopyranoside (coniferin) (**2**). Isolation and identification of phenylpropanoid derivatives from *A. tripedale* Trautv. is reported for the first time in this paper.

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## Introduction

*Allium* is the largest genus of the Amaryllidaceae family covering about 750 species including ornamental, edible and medicinal plants which are cultivated and collected all around the world with a more distribution in northern hemisphere [1-4].

Beside the use as common vegetables all over the world and also being an element of Mediterranean diet, *Allium* species are among medicinal plants which have been historically used for the treatment of many diseases specially cardiovascular diseases, hypercholesterolemia, obesity, diabetes, cataract, tumors and gastrointestinal tract complaints [2].

Phytochemically, *Allium* species are rich sources of important secondary metabolites involving steroidal saponin and sapogenins, flavonoids and organosulfur compounds [5,6] and numerous pharmacologically active compounds have been isolated from different *Allium species* in recent years. As a member of natural phenolic compounds, phenyl propanoid glycosides (PPGs) are valuable secondary metabolites widely distributed in different plant species including *Allium species* and have been demonstrated to possess different pharmacological activities specially in the treatment of hypertension, viral infections, fungal infections, tumors, cancer and immune system related disorders [7-11].

Honey garlic, *Allium tripedale* Trautv. is an spicy edible *Allium* of "Zagros" mountainous region in the west and northwest of Iran, wildy grows in this region and its strongly odorous leaves are used widely by local people as a raw vegetable and also to make different dishes [12]. The plant is locally called "Khargerju" or "Anashq" and beside the culinary uses is also considered as a medicinal plant, traditionally has been used for the treatment of various infectious conditions [13]. As a part of our research program on different *Allium* species, phytochemical investigation of *A. tripedale* has been conducted and the current paper reports the isolation and identification of the main phenolic constituents have been isolated from the chloroform-methanolic extract of the plant.

## Materials and Methods

### General experimental procedures

Medium pressure liquid chromatography (MPLC) was performed by a Buchi Gradient System C-605 apparatus using glass columns of LiChroprep® RP-18 (25-40µm) and C-660 Buchi fraction collector. TLC performed on SiO<sub>2</sub> plates with BuOH:H<sub>2</sub>O:CH<sub>3</sub>COOH (60:25:15 v/v/v) (BAW) as a mobile phase and cerium sulfate in 2N H<sub>2</sub>SO<sub>4</sub> and natural product (NP) as reagents for visualizing the spots.

HPLC was performed by Waters 515 apparatus equipped with a refractive index detector (Waters 2414) and UV detector (Waters 2487, λ : 254, 365 nm), using semipreparative C18 (Novapak®, 7.8 x 300 mm) and analytical C18 columns (Novapak®, 3.9 x 300 mm) in isocratic mode.

H and C NMR spectra recorded by Bruker 400MHz (H at 400 MHz and C at 100 MHz) spectrometer, using solvent signal for calibration (CD<sub>3</sub>OD: δH=3.31, δC=49.0). Distortionless enhancement by polarization transfer (DEPT) experiments was used to determine the multiplicities of C NMR resonances.

2D heteronuclear multiple bond correlation (HMBC), optimized for <sup>2-3</sup>J<sub>CH</sub> of 8 Hz, was used for determination of two and three bond heteronuclear<sup>1</sup>H-<sup>13</sup>C connectivities, while 2D Heteronuclear Single-Quantum Coherence (HSQC), interpulse delay set for <sup>1</sup>J<sub>CH</sub> of 130 Hz, was used for determination of one-bond heteronuclear<sup>1</sup>H-<sup>13</sup>C connectivities. ESIMS spectra were prepared by Shimadzu LCMS 2010 EV, using methanol as the solvent.

### Plant material

The whole plant of *A. tripedale* was collected from Khorram-Abad, Lorestan province on April 2013. The plant was identified by botanist and a voucher specimen (No. 3579) deposited at the Department of Pharmacognosy, Faculty of Pharmacy, Isfahan University of Medical Sciences, Iran.

### Extraction and Isolation

Air-dried underground parts of *A. tripedale* were finely powdered by means of a mill and the powder (500 g) was extracted at room temperature in a four step extraction method with increasing solvent polarity using the solvents; hexane, chloroform, chloroform-methanol (9:1) and methanol. Extraction was done using maceration method, performing each step four times with 2 L of solvent under occasional stirring. The chloroform-methanol (9:1) extract was concentrated under vacuum, yielding a dried extract (12 g) which was then fractionated by MPLC on a RP-18 column (36\*460 mm) using a linear gradient solvent system of H<sub>2</sub>O to MeOH. Fractions were analyzed by TLC (SiO<sub>2</sub>, BAW 60:15:25 v/v/v, reagents: cerium sulfate in 2N H<sub>2</sub>SO<sub>4</sub> and natural product (NP)) and similar fractions were mixed together.

Based on TLC and preliminary HNMR analysis, the fraction 3 was considered to be rich in phenolic compounds which after concentration by rotary evaporator, subjected to HPLC using a semi preparative C18 column (Novapak®, 7.8x300 mm) and H<sub>2</sub>O:CH<sub>3</sub>OH (65:35) mobile phase in isocratic mode, resulted the pure compounds (1) (10 mg) and (2) (16 mg).

### Results

Final purification of the main phenolic fraction resulted to the isolation of 2 pure phenyl propanoid glycosides. Structure elucidation of the isolated compounds were performed by comprehensive spectroscopic methods, resulted to the identification of coniferyl β-D-Glucopyranoside (Coniferin, Abietin) and 3-(4-hydroxyphenyl) propyl β-D-Glucopyranoside.

### Characterization of compounds (1)

ESIMS spectra of compound (1) in the negative-ion mode showed a pseudomolecular ion peak at m/z 313 [M-H], while the HNMR spectrum exhibited 2 characteristic aromatic proton signals (δ<sub>H</sub> 6.71 and δ<sub>H</sub> 7.08), 2 sp<sup>3</sup> proton signals (δ<sub>H</sub> 2.85

and δ<sub>H</sub> 3.20) and an anomeric proton signal (δ<sub>H</sub> 4.31) together with overlapped sp<sup>3</sup> proton signals (δ<sub>H</sub> 3.22-3.81), suggesting the glycosylated aromatic nature for compound (1).

CNMR spectral analysis of (1) showed 13 carbon signals including 2 double height sp<sup>2</sup> aromatic methine carbons (δ<sub>C</sub> 116.13 and 130.91), 2 quaternary sp<sup>2</sup> carbon signals (δ<sub>C</sub> 130.76, 156.82), an anomeric carbon signal (δ<sub>C</sub> 104.60) and 8 sp<sup>3</sup> carbon signals (δ<sub>C</sub> 19.80-78.25) which in agreement with MS and HNMR spectral data confirmed the glycosylated aromatic nature of compound (1) and suggested the molecular formula as C<sub>15</sub>H<sub>22</sub>O<sub>7</sub> (Table 2).

Finally by determination of low and high range <sup>1</sup>H-<sup>13</sup>C connectivities through HSQC and HMBC experiments, and comparing the spectral data with those reported for similar compounds in the literature [14-16], the assignments including the attachment of glucose residue to C1' were confirmed and the chemical structure of (1) was defined as 3-(4-hydroxy phenyl)propyl β-D-Glucopyranoside (Fig. 1).

### Characterization of compounds (2)

ESIMS spectra of compound (2) in the negative-ion mode showed a pseudomolecular ion peak at m/z 341 [M-H], while the HNMR spectrum exhibited the characteristic signals of aromatic protons (δ<sub>H</sub> 6.28 -7.14) as well as an isolated doublet at δ<sub>H</sub> 4.23 (2H, dd, J= 5.71, 1.26), an anomeric proton signal at δ<sub>H</sub> 4.91(1H, d, J= 7.37) and overlapped proton signals of sugar residue (δ<sub>H</sub> 3.33-3.79), altogether suggested the chemical structure of compound (2) as a glycosylated phenolic compound.

CNMR spectrum of compound (2) showed 16 carbon signals included those related to OCH<sub>3</sub> carbon (δ<sub>C</sub> 56.74), C2-C6 carbon signals of sugar residue (δ<sub>C</sub> 62.52-78.23), sp<sup>2</sup> carbon signals (δ<sub>C</sub> 111.42-150.91) and anomeric carbon signal (δ<sub>C</sub> 102.77), which in agreement with MS and HNMR spectral data confirmed the glycosylated phenolic nature of compound (2) and suggested the molecular formula as C<sub>16</sub>H<sub>22</sub>O<sub>8</sub> (Table 2).

Finally by determination of low and high range <sup>1</sup>H-<sup>13</sup>C connectivities through HSQC and HMBC

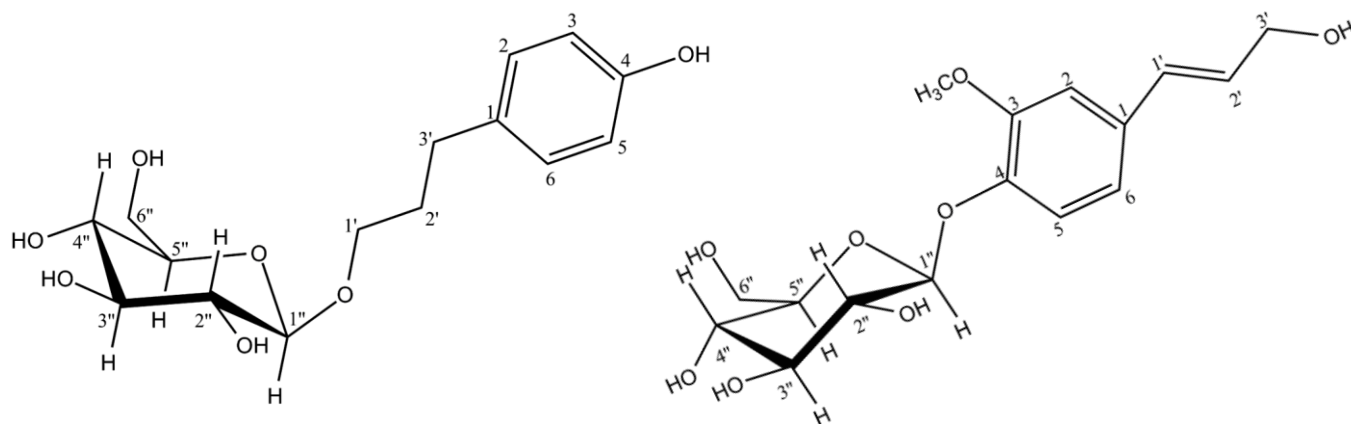
experiments, and comparing the spectral data with those reported for similar compounds in the literature [14,17-19], the assignments specially the attachment of OCH<sub>3</sub> to C3 and glucose residue to

C4 were confirmed and the chemical structure of (1) was defined as coniferyl β-D-Glucopyranoside (Coniferin, Abietin) (Fig. 2).

**Table 1.** <sup>1</sup>H and <sup>13</sup>CNMR data of compound (1) and (2) (400 MHz, 100 MHz, CD<sub>3</sub>OD)

Position	Compound (1)			Compound (2)		
	δ <sub>c</sub> (mult)	δ <sub>H</sub> (int ,mult, J)	HMBC (C)	δ <sub>c</sub> (mult)	δ <sub>H</sub> (int ,mult, J)	HMBC (C)
1	130.76 (C)	-	-	133.70 (C)	-	-
2	130.91(CH)	7.08 (1H,d,8.4)	3',3,6	111.42 (CH)	7.09 (1H,d,1.9)	1',3,6
3	116.13(CH)	6.71 (1H,d,8.4)	1,2,5	150.91 (C)	-	-
4	156.82(C)	-	-	147.65 (C)	-	-
5	116.13(CH)	6.71 (1H,d,8.4)	3', 2,3,6	117.96 (CH)	7.14(1H,d,8.4)	1',2,3,4,6
6	130.91 (CH)	7.08 (1H,d,8.4)	3',2,5	120.76 (CH)	6.97 (1H,dd,8.4,1.9)	1',2,5
1'	36.38 (CH <sub>2</sub> )	3.20 (2H,t,8)	2',3'	131.30 (CH)	6.56 (1H,d,15.8)	3',1,2,6
2'	19.80 (CH <sub>2</sub> )	0.95 (2H,dd,6.7,3.6)	1',3'	128.92 (CH)	6.28 (1H,dd,15.8,5.7)	1',3'
3'	71.68 (CH <sub>2</sub> )	2.85 (2H,t,8)	1',2,6	63.73 (CH <sub>2</sub> )	4.23 (2H,dd,5.71, 1.26)	1',2'
1''	104.60 (CH)	4.31 (1H,d,7.6)	2'', 1'	102.77 (CH)	4.91 (1H,d,7.37)	4,2''
2''	75.14 (CH)	3.48 (1H) <sup>a</sup>	1'',3''	74.92 (CH)	3.48 (1H) <sup>a</sup>	1'',3''
3''	77.98 (CH)	3.62(1H) <sup>a</sup>	2'',4'',5''	77.86 (CH)	3.60(1H) <sup>a</sup>	2'',4'',5''
4''	72.10 (CH)	3.22(1H) <sup>a</sup>	3'',5'',6''	71.36 (CH)	3.20(1H) <sup>a</sup>	3'',5'',6''
5''	78.25 (CH)	3.35(1H) <sup>a</sup>	4'',6''	78.23 (CH)	3.33(1H) <sup>a</sup>	4'',6''
6''	62.80 (CH <sub>2</sub> )	3.81(2H) <sup>a</sup>	4'',5''	62.52 (CH <sub>2</sub> )	3.79(2H) <sup>a</sup>	4'',5''
OCH <sub>3</sub>	-	-	-	56.74 (CH <sub>3</sub> )	3.90 (3H,s)	2,3,4

<sup>a</sup> overlapped with other signals



**Fig. 1.** Chemical structure of phenyl propanoid glycosides isolated from *A. tripedale*.

## Discussion

Phytochemical investigation of the edible plant, *Allium tripedale*, resulted to the isolation of two phenyl propanoid glycosides from the underground parts of the plant, using the comprehensive spectroscopic methods, were identified as 3-(4-hydroxyphenyl)propyl  $\beta$ -D-Glucopyranoside (**1**) and coniferyl  $\beta$ -D-Glucopyranoside (coniferin) (**2**).

Phenyl propanoid glycosides are widely distributed phenolic compounds which have been isolated from numerous plant species including *Allium* species [7, 16-19] and have been shown to possess many pharmacological effects especially in the treatment of hypertension, viral and fungal infections, cancer and immune related disorders. They also have been demonstrated to bear significant antioxidative effects, mainly by inhibition of low-density lipoproteins oxidation through free radical scavenging and ion chelating activity [7-11,20].

According to their antioxidant activity, PPGs have also been suggested to be valuable in the

prevention and treatment of a variety of diseases like cardiovascular and neurodegenerative disorders, atherosclerosis and diabetes type 2 [16, 20].

As a famous PPGs, Coniferin (Abietin) is an astringent glycoside which has been firstly isolated from coniferous plant species, while it has been isolated later from a variety of plant species mainly some *citrus* and *asparagus* species, comfrey and sugar beet. Beside the biological role as a lignin precursor in many plant species, coniferin has been reported to show different pharmacological effects including anti-cough and anti-asthmatic properties [21]. Although different phenylpropanoid derivatives have been isolated from a vast range of plant families including Amaryllidaceae [7,16-19], there is just a report about the isolation and identification of ferulic acid-4-O- $\beta$ -D-glucopyranoside from the underground parts of *Allium hookeri* [17], and to the best of our knowledge isolation of 3-(4-hydroxyphenyl)propyl  $\beta$ -D-glucopyranoside (**1**) and coniferyl  $\beta$ -D-glucopyranoside (coniferin) (**2**) from *Allium* species is reported for the first time in this study. The results could be used for

explanation of medicinal activities attributed to *A. tripedale* or to candidate the phenolic reached extract of the plant for new pharmacological studies based on the biological and pharmacological effects observed from different PPGs.

## Conclusion

Phenyl propanoid glycosides are important secondary metabolites which have been reported to possess different pharmacological activities. Isolation and identification of phenyl propanoid glycosides from the underground parts of *A. tripedale* is reported for the first time in this study and could be used both as a chemical basis for explanation of medicinal effects of the plant and conducting pharmacological studies based on the medicinal activities of Phenyl propanoid glycosides.

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## Conflict of interest

Authors certify that there is no actual or potential conflict of interest in relation to this article.

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