

The Comparison of the Effect of Diazinon and Paraoxon on Biomarkers of Oxidative Stress in Rat Serum

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Abstract

Background: Diazinon and paraoxon are a group of pesticide organophosphates which are widely used in agriculture. Some organophosphates increase the production of free radicals. This study aims to compare the effects of diazinon and paraoxon on biomarkers of oxidative stress in rat serum.

Materials and Methods : This experimental study was performed in 2010 in Baqiyatallah (a.s) University of Medical Sciences. 49 male Wistar rats were randomly divided into 7 groups: the control group was given corn oil as a solvent of diazinon and paraoxon and the other six groups were given diazinon at doses 30, 50 and 100 mg/kg and paraoxon at doses 0.3, 0.7 and 1 mg/kg intraperitoneally. 24 hours after the injection, blood was collected by cardiac puncture and serum was prepared. Then, the activities of butyrylcholinesterase enzyme (BChE), superoxide dismutase (SOD), catalase (CAT) and lactate dehydrogenase (LDH) and glutathione level (GSH) were measured by chemical methods.

Results: The decreased SOD and CAT activities and GSH level after administration of Diazinon (more than 30 mg/kg) and paraoxon (more than 0.3 mg/kg) is significant compared to control group, whereas the activity of LDH enzyme significantly increases. The decreased BChE activity is not significant compared to control group.

Conclusion: Diazinon and paraoxon induce the production of free radicals and oxidative stress in a dose-dependent manner. The decreased activity of antioxidant enzymes and depleted GSH content probably represent the antioxidant defense system failure against the toxic actions of free radicals and oxidative tissue damage. Paraoxon has more severe effects than diazinon.

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Introduction

Organophosphate compounds are widely used to control insects and ectoparasites. In addition, these compounds including tabon, sarin and malathion have been used in several chemical attacks by Iraqi forces during Iran-Iraq war. Due to the easy access and high toxicity of these compounds, the incidence of accidental poisoning and suicide is extensive and it is responsible for about one hundred thousand annual poisoning in the world. In Iran, these compounds are one of the reasons for deaths from poisoning [1, 2]. Diazinon (DZN) and paraoxon (POX) are of main organophosphate pesticides which are used for the control of various insects in crops, ornamental plants, fruit and vegetables. These compounds with amino acid serine phosphorylation in the active site of acetylcholinesterase enzyme leads to the inhibition of the enzyme and accumulation of acetylcholine in cholinergic synapses and cholinergic crisis, seizures and in acute cases, brain injury and death [2, 3]. Being

under acute and sub-acute exposure of some organophosphates inhibits cholinesterase and increases the production of free radicals and oxidative stress.² The reaction of free radicals with lipids in cell membranes causes lipid peroxidation and destruction of membrane structure. Harmful effects of free radicals is controlled by the cellular antioxidant defense system includes antioxidant enzymes such as superoxide dismutase (SOD), catalase (CAT). In addition, glutathione (GSH), as an antioxidant, increases the solubility and excretion of toxins through the kidneys. The imbalance between production of free radicals and antioxidant defense systems in the body is called oxidative stress, which causes many diseases [4, 5]. Studies showed that after administration of oral malathion to rats, changes of activity of antioxidant enzymes and increase of lipid peroxidation are observed in erythrocyte, saliva and plasma [2, 6]. On the other hand, by examination of the effect of chlorpyrifos and cypermethrin on liver

rat increased serum concentrations of liver function indices and changes in antioxidant parameters have been reported [7]. Due to the variation of substitutions in chemical structure of organophosphates and their different effects on different tissues, further studies are necessary to understand the mechanism of action of these compounds. Very few studies are conducted on the effects of diazinon on in vivo antioxidant system. In the present study, the effects of two toxins of DZN and POX on some indices of oxidative stress in rat serum have been compared.

Materials and Methods

Chemicals: Nitroblue tetrazolium (NBT), dithiobis 2-nitrobenzoic acid (DTNB), trichloroacetic acid (TCA) and other chemicals with high purity were purchased from Merck and Sigma Companies of Germany. Diazinon (purity of 100%) from Supelco-USA and ethyl paraoxon (purity of 99%) from Sigma Company were purchased. Stock solutions of DZN with concentration of 400 mg/ml and POX with concentration of 4 mg/ml were prepared freshly in corn oil.

This experimental study was performed in 2010 in Baqiyatallah (a.s) University of Medical Sciences. 49 male Wistar rats weighing 200-250 g were kept under natural conditions of light, darkness, food and water for a week to get accustomed to the environment in animal house of Baqiyatallah University of Medical Sciences. The animals were then randomly divided into 7 groups: the control group was given corn oil as a solvent and the other six groups were given DZN (at doses 30, 50 and 100 mg/kg) and POX (at doses 0.3, 0.7 and 1 mg/kg) intraperitoneally. 24 hours after the injection, blood was collected by cardiac puncture and serum was prepared and stored at -70°C until biochemical analysis.

Butrylcholinesterase activity measurement: Ellman method was used to measure the activity of cholinesterase [8]. 10 μl of serum was added to the cuvet containing 0.425 mM DTNB dissolved in 0.1 M sodium phosphate buffer pH 7.5 with 490 μl of distilled water. Then, 0.5 ml of butrylthiocholine iodide of 20mM as substrate was added to the cuvet. It was incubated at 37°C for 5 minutes. Then, 0.5 ml Quinidine sulfate was added to stop the reaction. The absorbance was read at wavelength of 412 nm for 5 minutes. Specific activity was calculated based on the unit/ mg protein.

Measurement of SOD activity: SOD enzyme activity was measured using Winterbourn method [9]. 0.1 M EDTA in 0.3 mM sodium cyanide and 1.5 mM NBT were added to an appropriate volume of serum in a cuvet and after mixing, it was incubated at 37°C for 5 minutes. Then, 0.12 mM riboflavin was added to 0.067 M potassium phosphate buffer with pH=7.8 and it was placed at room temperature for 10 minutes. The absorbance was read at a wavelength of 560 nm for 5

minutes and the specific activity was calculated by the unit per mg protein.

Measurement of CAT activity: The activity of CAT enzyme was measured by Aebi method [10]. The reaction was started by adding 30 mM H_2O_2 to an appropriate volume of serum in 50 mM sodium phosphate buffer with pH=7. Then, the absorbance was read at a wavelength of 240 nm within 3 minutes. Specific activity was calculated based on the unit/mg protein.

Measurement of LDH activity: LDH enzyme activity was measured using Parsazmun Company kit of Iran. Sample absorbance at 340 nm was read within 3 minutes, and the specific activity was expressed by the unit per mg protein.

Determination of GSH concentration: Thietz method was used to determine GSH level [11]. An appropriate volume of serum was mixed with 5% Sulfosalicylic acid. The sample was centrifuged at $2000 \times g$ for 10 min at 4°C . 100 μl of supernatant was added to 810 μl of 0.3 M disodium phosphate. Then the reaction was started by adding 90 μl of 0.04% DTNB in 0.1% sodium citrate. The absorbance changes at 412 nm were read within 5 minutes. Standard curves were plotted and glutathione concentration was calculated using 1mg/ml glutathione solution.

Protein assay: Bradford method was used to determine protein concentration [12]. Appropriate volume of serum reached the volume of 1ml to which 3ml of Bradford solution was added and was incubated for 10 minutes. Then, absorbance was read at 595 nm wavelength. Protein concentration was calculated using standard plotting using 1mg/ml bovine serum albumin solution (BSA).

Statistical analysis: Data analysis was performed using INSTAT software as one-way ANOVA followed by Tukey post-hoc multiple comparison test. *p*-values less than 0.05 were considered statistically significant. Data were expressed as Mean \pm SD.

Results

The results of the effect of different doses of DZN and POX on the activity of serum butrylcholinesterase enzyme in Fig. 1 shows that the change of butrylcholinesterase enzyme activity is not significant compared to the control group.

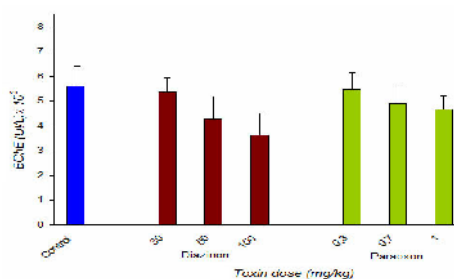


Figure 1. The activity of serum butyrylcholinesterase enzyme of rats treated with various doses of diazinon and paraoxon after 24 hours.

The results of the effect of different doses of DZN and POX on the activity of SOD in serum in Fig. 2 shows that the decrease in SOD activity at doses higher than 30 mg/kg DZN and 0.3 mg/kg POX is significant compared to the control group. In addition, decreased activity of SOD at dose 100 mg/kg is significant compared to dose 30 mg/kg of DZN ($p < 0.01$).

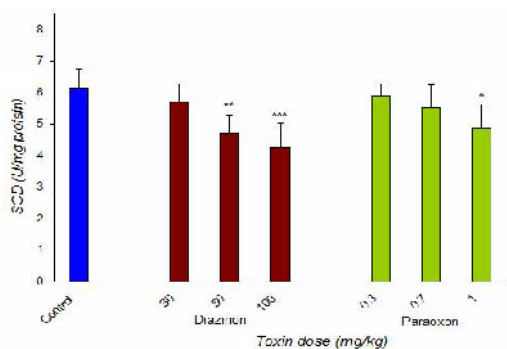


Figure 2. The activity of superoxide dismutase in serum after 24-hour at presence of different doses of diazinon and paraoxon. * $p < 0.05$, ** $p < 0.01$ and *** $p < 0.001$ vs. control.

The results of the effect of different doses of DZN and POX on the activity of CAT of serum in Fig. 3 shows that the decrease in CAT activity at doses higher than 30 mg/kg DZN and 0.3 mg/kg POX is significant compared to the control group. In addition, CAT activity at dose 1mg/kg significantly decreases compared to dose 0.3 mg/kg of POX ($p < 0.05$).

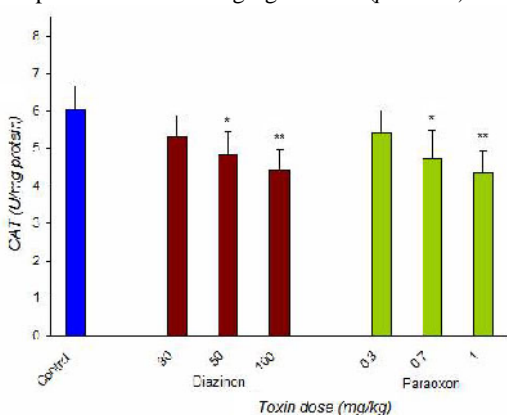


Figure 3. Serum catalase activity enzyme of rats treated with various doses of diazinon and paraoxon 24 hours after exposure. * $p < 0.05$ and ** $p < 0.01$ vs. control.

The results of the effect of different doses of DZN and POX on the activity of LDH in serum in figure 4 shows that the increase in LDH activity at doses higher than 30 mg/kg DZN and 0.7 mg/kg POX is significant compared to the control group. The LDH enzyme activity in dose 1 mg/kg significantly

increases compared to dose 0.3 mg/kg of POX ($p < 0.01$).

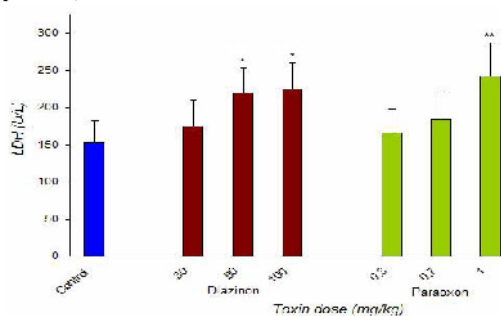


Figure 4. Effects of various doses of diazinon and paraoxon on activity of serum LDH enzyme after 24. * $p < 0.05$ and ** $p < 0.01$ vs. control.

The results of the effect of different doses of DZN and POX on GSH concentration of serum in Fig. 5 shows that decrease of GSH concentration at doses higher than 30 mg/kg DZN and 0.3mg/kg POX is significant compared to the control group. In addition, GSH concentration in dose 100 mg/kg significantly decreases compared to dose 30 mg/kg of DZN ($p < 0.01$) and dose 1mg/kg significantly decreases compared to dose 0.3 mg/kg of POX ($p < 0.01$).

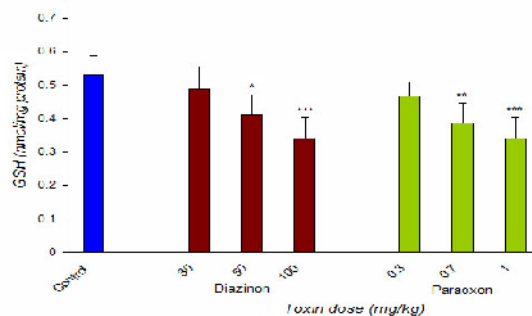


Figure 5: The glutathione concentration in serum after 24-hour at presence of different doses of diazinon and paraoxon. * $p < 0.05$, ** $p < 0.01$ and *** $p < 0.001$ vs. control

Discussion

The results of this study indicate that the activity of enzymes SOD and CAT decreases at doses higher than 30 mg/kg DZN and 0.3 mg/kg POX. Serum contains little CAT, SOD, GPx and GSH; its antioxidant properties are dependent on small molecules such as ascorbic acid, bilirubin, uric acid, ferritin and transferrin [13]. SOD is the first line of defense against superoxide anion radicals, which converts it into H_2O_2 and O_2 . H_2O_2 is converted into H_2O and O_2 through CAT enzyme. The decrease in SOD activity probably increases superoxide radicals. This increase had an inhibitory effect on CAT activity and decrease CAT activity, which can lead to tissue damage and oxidative stress [4]. This study results are consistent with results of studies showing that

organophosphate of phosphamidon and DZN inhibit the activity of SOD [14, 15]. Khan et al with study on the effect of chlorpyrifos and cypermethrin (administration for 2 weeks) on liver rats reported that the activity of SOD and CAT were decreased.⁷ On the other hand, in some studies increased activity of antioxidant enzymes has been reported after prescription of the pesticide compounds. Sharma et al showed that administration of oral dimethoate to rats causes a dose-dependent increase in SOD and CAT activity in brain and liver [16].

The studies of Altuntas showed the increase of SOD activity during DZN and Fenton poisoning [17]. The studies of Monterio et al showed that methyl parathion or folisuper leads to the production of free radical and increased activity of SOD and CAT enzyme in liver and muscle of fish [18]. The studies of Akturk et al showed that DZN increased the activity of SOD and CAT in rat heart and RBC [19]. The studies of Ghani et al showed that intraperitoneal administration of POX for 4 hours in rats increased the activity of enzymes SOD and CAT in the brain tissue [20]. The studies of Isik et al showed that consumption of DZN and methyl parathion to fish for 24 hours increases the activity of SOD in liver and lungs [21]. This difference of results in different studies is caused by the animal type, breed and species, tissue type, dose, exposure time and administration rout of toxin.

LDH is a cytoplasmic enzyme which is used to evaluate cellular damage as a marker to examine toxicity of a chemical substance and cell lysis. The activity of this enzyme is directly correlated with the extent of cell death [22]. In this study, the effect of DZN and POX on serum LDH activity shows that LDH activity increases at doses higher than 30 mg/kg DZN and 1 mg/kg POX. The increase of this enzyme may be due to the increased tissue damage and its leak into the serum. The studies of Gokcimen et al showed that DZN administration to rats for 24 hours increased the activity of LDH enzyme in the liver and pancreas [23]. In animal studies, Manna et al also reported the increased activity of serum LDH after administration of some compounds such as alpha-cypermethrin [24]. Mishra et al showed that consumption of endosulfan to fish decreased LDH activity in the liver and skeletal muscle [25]. The studies of Ghani et al showed that administration of POX to rats for 4 hours decreased LDH of liver [20]. Different results could be due to differences in administration type, the concentration of the toxin used, breed and species of animal, tissue type and even specific LDH isoenzymes.

GSH is a peptide containing thiol, one of main cellular antioxidant which is synthesized in the liver cytoplasm and then distributed to the other tissues through blood. GSH plays an important role in scavenging ROS and detoxification of drugs and chemicals [26]. GSH is essential for conversion of dehydroascorbic into ascorbate which acts as antioxidant to neutralize superoxide [27]. In the

present study, the concentration of serum GSH decreases at doses higher than 30 mg/kg DZN and 0.3 mg/kg POX. The decreased GSH may be due to increased utilization of GSH for conjugation and/or participation of GSH as an antioxidant in neutralizing free radicals [6].

In addition, decreased GSH of serum is also a reflection of hepatic GSH synthesis reduction which decreases GSH of extrahepatic tissues and disturbs GSH homeostasis of the whole-body [4, 5]. The studies of Buyukokuroglu et al showed that, Fenthion consumption decreased GSH level of human muscle after 24 hour [28].

The studies of Monteiro et al also showed that the consumption of methyl parathion (MP) by fish decreased GSH level in tissues of liver, white muscle and gills [18]. The studies of Khan et al showed that consumption of cypermethrin and malathion by rats reduces GSH level in the liver [7].

Butrylcholinesterase is one of the target enzymes of organophosphates (diazinon and paraoxon) whose toxic effects are created through cholinesterase inhibition, which is necessary for nervous system function. This enzyme exists in serum and liver and is particularly important in terms of pharmacology and toxicology and quickly reveals the acute effects of organophosphates [2, 29].

Measuring the activity of this enzyme has high value in the diagnosis of poisoning with insecticides. In this study, serum BChE activity did not change significantly due to the administration of DZN and POX. This result is probably due to the improvement of serum BChE activity after 24 hours and performance of other body systems such as antioxidant defense system. In addition, the effect of organophosphates on BChE activity is dependent on the consumed dose, contact duration, and type of tissue and animal. The studies of Sun et al showed that BChE enzyme activity makes no changes in the blood sample after prescription of human BChE and then, administration of soman after 14 months [30]. The studies of Sayim et al showed that BChE enzyme activity was decreased after consumption of dimethoate to rats for 90 consecutive days [31].

The studies of Nikookar et al showed that after consumption of malathion for 14 days to rats reduces the activity of BChE in the liver and serum [32]. Comparison of the effect of DZN and POX on the parameters of serum oxidative stress shows that both toxins act similarly. However, the POX dose used, which is much less than DZN, indicates its greater toxicity. P=S bond exists in diazinon structure, which turns into P=O bond in paraoxon structure, This change leads to more severe activity of this organophosphate [33, 34].

In summary, the results suggest that the effects of DZN and POX on serum antioxidant defense system are in a dose-dependent manner. The decreased activity of antioxidant enzymes and depleted GSH

content probably represents the antioxidant defense system failure against the toxic actions of free radicals, which may lead to oxidative stress. POX has much more severe effects.

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