



Design, Fabrication, and Evaluation of Solid Lipid Nanoparticles Containing *Ferula assa-foetida* Essential Oil as an Antimicrobial Agent Against Human Pathogens

Mehdi Jahantigh ¹, Saeide Saeidi ^{2,*}, Masoomah Rezaei ³

¹ Department of Clinical Sciences, Faculty of Veterinary Medicine, University of Zabol, Zabol, Iran

² Biotechnology Research Institute, National University of Zabol, Zabol, Iran

³ Cellular and Molecular Research Institute, Zahedan University of Medical Sciences, Zahedan, Iran

*Corresponding Author: Biotechnology Research Institute, National University of Zabol, Zabol, Iran. Email: s.saeedi12@yahoo.com

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Abstract

Background: *Ferula assa-foetida* essential oil has demonstrated antimicrobial activity, making it a promising candidate for plant-derived antimicrobial formulations. However, poor aqueous solubility, volatility, and instability may limit the application of essential oils. Encapsulation in solid lipid nanoparticles (SLNs) may help overcome these limitations and improve antimicrobial delivery. This study aimed to formulate and characterize SLNs containing *F. assa-foetida* essential oil and to evaluate their antibacterial activity against foodborne pathogens.

Objectives: This study aimed to design, fabricate, and evaluate solid lipid nanoparticles containing *F. assa-foetida* essential oil as an antimicrobial agent against foodborne pathogens.

Methods: Lipids were heated to 5 - 10 °C above their melting point. Lipophilic substances were added to the lipid phase, and hydrophilic substances were added to the aqueous phase. Both phases were maintained at the same temperature in a water bath. The aqueous phase was added to the lipid phase to form a pre-emulsion, which was homogenized using an Ultratorex homogenizer and then reduced to fine particles using an ultrasound device. SLN formation at room temperature required approximately 4 - 12 hours and, in some cases, up to 24 hours. Fourier transform infrared spectroscopy (FTIR) and electron microscopy analyses were then performed, and incubation rates were assessed. The antimicrobial properties of the SLNs were assessed using antibacterial methods.

Results: The FTIR spectrum of SLNs containing *F. assa-foetida* essential oil showed functional groups associated with the lipid matrix and the active compounds of the essential oil, including phenolic and alcoholic compounds and hydrogen-bonding interactions. The differential thermal analysis (DTA) plot showed three main thermal events. A small change at approximately 25.5 °C, with a negative heat change, was attributed to the release of moisture or volatile compounds. The event at 262.2 °C could reflect the onset of decomposition of light organic compounds and plant components of the *F. assa-foetida* essential oil. The main broad peak began at 485.5 °C and peaked at approximately 560 - 570 °C, corresponding to major decomposition of organic compounds and the lipid matrix. Overall, DTA indicated relative stability up to intermediate temperatures and major degradation at high temperatures. The SLNs produced inhibition halos and showed good inhibitory activity against foodborne pathogenic bacteria.

Conclusions: Physicochemical evaluation of the nanosystem after loading *F. assa-foetida* essential oil into SLNs showed that the system was anionic and exhibited slow-release behavior. Its particle size, appropriate morphology, and acceptable essential oil uptake rate indicated its suitability as a carrier for *F. assa-foetida* essential oil, thereby increasing the essential oil's stability and solubility and consequently improving its performance.

Keywords: Lipid Nanoparticles, *Ferula Assa-foetida* Essential Oil, Antimicrobial, Foodborne Pathogens

1. Background

Medicinal plants have been used since ancient times to treat diseases. Their use worldwide, particularly in developing countries such as India, contributes substantially to primary health care. Herbal medicines have considerable potential that has not been adequately explored. Knowledge of medicinal plants is sometimes considered the only source of treatment for certain communities and ethnic groups.

Essential oils contain volatile compounds, including terpenoids, phenols, and aliphatic components, and are widely used in the food and pharmaceutical industries (1).

Because essential oils are highly volatile, encapsulation methods can increase their aqueous solubility, reduce volatility, and ultimately improve their effectiveness (2).

Nanocarriers can be produced using a wide range of materials and designs (3). Liposomes are bilayer vesicles formed by the assembly of amphiphilic lipids (4-5).

New drug delivery methods using nanotechnology encapsulate drugs in carrier systems to increase drug stability and solubility (6-7).

Newer methods use lipid viscogels, including flexible liposomes with shape-changing properties, and liposomes containing propylene glycol or polysorbate (8).

The genus *Ferula* belongs to the family Apiaceae (9). *Ferula* species exhibit analgesic, anti-inflammatory, antipyretic, antispasmodic, and anticonvulsant effects (10-11).

Various studies have demonstrated the antimicrobial and antifungal activities of these plants (12).

Previous studies have also shown that the essential oil of this plant contains sulfur compounds and monoterpenes (13).

2. Objectives

This study aimed to design, fabricate, and evaluate solid lipid nanoparticles containing *Ferula assa-foetida* essential oil as an antimicrobial agent against food-borne pathogens.

3. Methods

3.1. Formulation Design and Data Analysis

Formulations were designed using Design Expert software and the response surface method. This software was used to select the optimal formulation while reducing time and cost. A central composite design was used to design the formulations. The variables included the percentage of lipid (stearic acid), surfactant (Poloxamer 188 and Span 80), and essential oil, as well as the Poloxamer 188:Span 80 ratio. Variable ranges were selected based on published studies: 3% - 5% for stearic acid, 1% - 3% for *F. assa-foetida* essential oil, 1% - 2% for surfactant, and 0.5 - 2 for the dimensionless Poloxamer 188:Span 80 ratio.

3.2. Preparation of Formulations by Hot Homogenization and Ultrasound

Lipids were heated to 5 - 10 °C above their melting point. Lipophilic substances were added to the lipid phase, and hydrophilic substances were added to the aqueous phase. Both phases were maintained at the same temperature in a water bath. The aqueous phase was added to the lipid phase to form a pre-emulsion, which was homogenized using an Ultratorex homogenizer and then reduced to small particles using an ultrasound device. The formation of solid lipid nanoparticles (SLNs) at room temperature required approximately 4 - 12 hours and, in some cases, up to 24 hours.

3.3. Nanoparticle Size

Electron microscopy and a particle size analyzer were used to assess the size of the prepared SLNs. Particle size and zeta potential were measured.

3.4. Physical Stability

Color and odor were observed and recorded after 1 month and 6 months at room temperature.

3.5. Nanoparticle Morphology

Nanoparticle morphology was examined using scanning electron microscopy (SEM) and transmission electron microscopy (TEM). Samples were diluted with double-distilled water and placed on a carbon-coated copper grid. They were negatively stained with 1% phosphotungstic acid, dried at room temperature, and observed using a microscope (14).

3.6. Fourier Transform Infrared Spectroscopy

Fourier transform infrared spectroscopy (FTIR) was used to observe infrared peaks arising from changes in molecular vibrational energy levels. First, 10 mg of the sample was placed in a mortar with 50 times its amount of potassium bromide (KBr) and ground counterclockwise with a pestle until a uniform mixture was obtained. The pestle was then placed inside the die through the transparent part. The uniform powder was placed inside the die and smoothed with a pestle. Another pestle was placed on top, and the steel cylinder was placed inside the die. The assembly was pressed using a press machine at a load of 10 tons for 2 minutes. After the die was opened, the resulting disk was placed on its base and transferred to the instrument. Results were recorded using Spectrum software.

3.7. Minimum Inhibitory and Minimum Bactericidal Concentrations

Gentamicin is often used as a reference antibiotic in microbial tests because it is a broad-spectrum antibiotic effective against both gram-positive and gram-negative bacteria.

Bacterial strains were obtained from a standard laboratory. The antibacterial activity of the SLNs was investigated against *Staphylococcus aureus* ATCC 1189, *Listeria monocytogenes* ATCC 1298, and *Bacillus cereus* PTCC 1015. The minimum inhibitory concentration (MIC) and minimum bactericidal concentration (MBC) were evaluated.

3.8. Statistical Analysis

Data were subjected to one-way analysis of variance (ANOVA) using SPSS version 17.

4. Results

4.1. Fourier Transform Infrared Spectroscopy

The FTIR spectrum of SLNs containing *F. assa-foetida* essential oil showed functional groups associated with the lipid matrix and the active compounds of the essential oil. The broad band at 3200 - 3600 cm^{-1} was attributed to O-H stretching vibrations, which may arise from phenolic and alcoholic compounds in the essential oil, as well as hydrogen-bonding interactions. Peaks at approximately 2920 and 2850 cm^{-1} corresponded to C-H stretching vibrations in lipid aliphatic chains, confirming the presence of a lipid phase in the nanoparticle structure. The band at 1700 - 1750 cm^{-1} corresponded to stretching vibrations of the carbonyl group (C=O), indicating acidic or ester compounds in the formulation. Bands at 1600 - 1500 cm^{-1} were attributed to C=C vibrations, and those at 1000 - 1300 cm^{-1} to C-O and C-O-C vibrations, indicating oxygenated and bioactive compounds in *F. assa-foetida* essential oil. Overall, the FTIR results indicated successful loading of the essential oil into the lipid nanoparticle matrix. The absence of new peaks or significant band shifts indicated that interactions between the essential oil and lipids were mainly physical, with no significant chemical reaction (Figure 1).

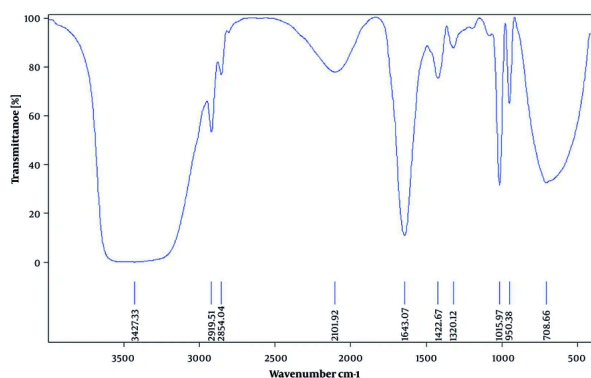


Figure 1. Fourier transform infrared spectroscopy (FTIR) of solid lipid nanoparticles

4.2. Differential Thermal Analysis

The differential thermal analysis (DTA) plot of SLNs containing *F. assa-foetida* essential oil showed three main thermal events. A small change at approximately 25.5 °C, with a negative heat change, was attributed to the release of moisture or volatile compounds. The event at 262.2 °C may reflect the onset of decomposition of light organic compounds and plant components of *F. assa-foetida* essential oil. The main broad peak began at 485.5 °C and peaked at approximately 560 - 570 °C, corresponding to major decomposition of organic compounds and the lipid matrix. Overall, DTA showed relative stability up to intermediate temperatures and major degradation at high temperatures (Figure 2).

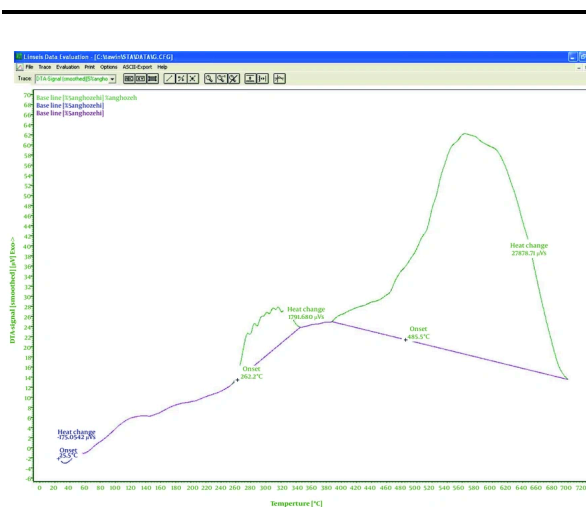


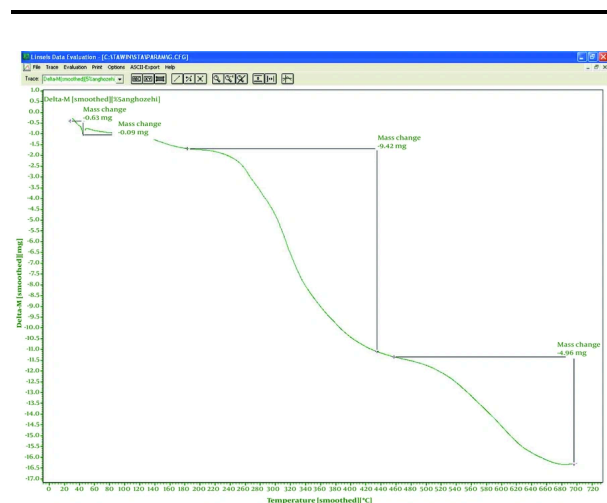
Figure 2. Differential thermal analysis (DTA) of solid lipid nanoparticles

4.3. Thermogravimetric Analysis

The thermogravimetric analysis (TGA) curve of SLNs containing *F. assa-foetida* essential oil showed thermal degradation in multiple stages. Initial weight loss at low temperatures was attributed to evaporation of absorbed moisture and the release of volatile compounds. Subsequent weight loss at intermediate temperatures may reflect decomposition of light organic compounds and bioactive components in the essential oil. The main weight loss at higher temperatures corresponded to major degradation of the lipid matrix and decomposition of organic chains. Residual weight at the end of the test was also likely attributable to ash, minerals, or thermally stable components. Overall, TGA showed relative thermal stability up to intermediate temperatures and main degradation at higher temperatures (Figure 3).

Table 1. Minimum Inhibitory Concentration (MIC) and Minimum Bactericidal Concentration (MBC) of Solid Lipid Nanoparticles (mg/mL)

Bacterial species	MIC	MBC
<i>Staphylococcus aureus</i>	50	100
<i>Listeria monocytogenes</i>	100	200
<i>Bacillus cereus</i>	100	200

**Figure 3.** Thermogravimetric analysis (TGA) of solid lipid nanoparticles

4.4. Antibacterial Activity

Antimicrobial testing showed that the lowest inhibitory concentration against *S. aureus* was 50 mg/mL, whereas the highest inhibitory concentration against *L. monocytogenes* was 100 mg/mL.

The lowest inhibitory concentration against *S. aureus* was 50 mg/mL. The highest inhibitory concentration was 100 mg/mL, which inhibited *L. monocytogenes* and *B. cereus* strains (Table 1).

5. Discussion

In a study examining the antimicrobial properties of *F. assa-foetida*, the hydroalcoholic extract inhibited *S. aureus* and *Bacillus subtilis* (15).

Majdizadeh et al. synthesized nanoliposomes containing peppermint essential oil and investigated their stability and antimicrobial, antioxidant, and antiproliferative properties. Formulation F1 was selected, with an encapsulation efficiency of 62.12%, a nanoparticle size of 121 nm, and a zeta potential of -21.8 mV (16).

Faraji et al. investigated the encapsulation of *Mentha pulegium* and *Ferula gummosa* essential oils using nanoliposome technology as a safe botanical pesticide. The mean ($\pm$ standard deviation) particle sizes of nanoliposomes containing *M. pulegium* and *F. gummosa* essential oils were 345 ± 3.2 and 309 ± 1.67 nm, respectively, and their encapsulation efficiencies were $99.38 \pm 0.24\%$ and $96.41 \pm 0.26\%$, respectively. The type of essential oil did not significantly affect the physicochemical properties of the nanoparticles. After 24 hours, essential oil release from *M. pulegium* and *F. gummosa* nanoliposomes was 46% and 33%, respectively. The estimated LC₅₀ values for nanoliposomes and crude *M. pulegium* essential oil against *T. castaneum* were 36.53 and 75.23 μ L/L air, respectively (17).

In a study preparing nanoliposomes with rosemary essential oil, the particle size was 162 nm, and the particles were spherical (18).

A study examining the combined activity of several essential oils against *Candida albicans* reported an inhibition halo diameter of 9 mm (19).

The effect of a chitosan nanoparticle biofilm containing *F. assa-foetida* on the healing of full-thickness wounds infected with methicillin-resistant *S. aureus* was investigated. Microbiological, planimetric, biomechanical, histological, and quantitative morphometric studies, along with the determination of hydroxyproline levels, showed a significant difference between animals in group 4 and those in the other groups (20).

A study investigating the antimicrobial activity of several medicinal plants against *Escherichia coli* showed that *P. harmala* extract had the highest antimicrobial activity, with a MIC of 2.5 mg/mL (21).

In a study investigating the synthesis of silver nanoparticles using *Sesamum indicum*, the minimum inhibitory concentration against *E. coli* was 200 ppm (22).

Studies synthesizing silver nanoparticles using *Plantago ovata* via a green method reported a nanoparticle size of 13 nm. The highest and lowest inhibitory concentrations were 100 and 12.5 mg/mL, respectively (23).

5.1. Conclusions

The results indicated that *F. assa-foetida* essential oil in SLNs had a long shelf life at 4 °C and exhibited good

antimicrobial properties.

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Footnotes

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